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The attached documents are exact copies of the European patent application described on the following page, as originally filed.

Les documents fixés à cette attestation sont conformes à la version initialement déposée de la demande de brevet européen spécifiée à la page suivante.



**Patentanmeldung Nr. | Patent application No. | Demande de brevet n°**

03104810.1

Der Präsident des Europäischen Patentamts;
Im Auftrag

For the President of the European Patent Office

Le Président de l'Office européen des brevets
p.o.

R C van Dijk

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Demande no:

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Anmelder/Applicant(s)/Demandeur(s):

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Bezeichnung der Erfindung/Title of the invention/Titre de l'invention:
(Falls die Bezeichnung der Erfindung nicht angegeben ist, siehe Beschreibung.
If no title is shown please refer to the description.
Si aucun titre n'est indiqué se referer à la description.)

MORPHOLINYL CONTAINING BENZIMIDAZOLES AS INHIBITORS OF RESPIRATORY SYNCYTIAL VIRUS REPLICATION

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# MORPHOLINYL CONTAINING BENZIMIDAZOLES AS INHIBITORS OF RESPIRATORY SYNCYTIAL VIRUS REPLICATION

The present invention is concerned with morpholinyl containing benzimidazoles having antiviral activity, in particular, they have an inhibitory activity on the replication of the respiratory syncytial virus (RSV). It further concerns their preparation and compositions comprising them, as well as their use as a medicine.

Human RSV or Respiratory Syncytial Virus is a large RNA virus, member of the family of Paramyxoviridae, subfamily pneumovirinae together with bovine RSV virus. Human RSV is responsible for a spectrum of respiratory tract diseases in people of all ages throughout the world. It is the major cause of lower respiratory tract illness during infancy and childhood. Over half of all infants encounter RSV in their first year of life, and almost all within their first two years. The infection in young children can cause lung damage that persists for years and may contribute to chronic lung disease in later life (chronic wheezing, asthma). Older children and adults often suffer from a (bad) common cold upon RSV infection. In old age, susceptibility again increases, and RSV has been implicated in a number of outbreaks of pneumonia in the aged resulting in significant mortality.

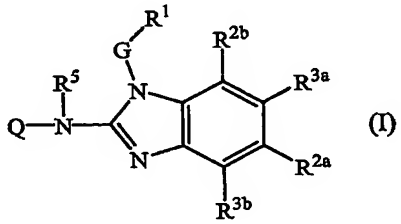
Infection with a virus from a given subgroup does not protect against a subsequent infection with an RSV isolate from the same subgroup in the following winter season. Re-infection with RSV is thus common, despite the existence of only two subtypes, A and B.

Today only three drugs have been approved for use against RSV infection. A first one is ribavirin, a nucleoside analogue, provides an aerosol treatment for serious RSV infection in hospitalized children. The aerosol route of administration, the toxicity (risk of teratogenicity), the cost and the highly variable efficacy limit its use. The other two drugs, RespiGam® and palivizumab, polyclonal and monoclonal antibody immunostimulants, are intended to be used in a preventive way.

Other attempts to develop a safe and effective RSV vaccine have all met with failure thus far. Inactivated vaccines failed to protect against disease, and in fact in some cases enhanced disease during subsequent infection. Life attenuated vaccines have been tried with limited success. Clearly there is a need for an efficacious non-toxic and easy to administer drug against RSV replication.

Previously, benzimidazoles and imidazopyridines as inhibitors of RSV replication have been described in WO 01/00611, WO 01/00612 and WO 01/00615.

The present invention now concerns inhibitors of RSV replication and are defined as the compounds of formula (I)

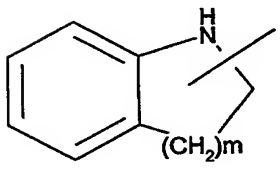


(I)

their prodrugs, *N*-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms wherein

G is a direct bond or $C_{1-10}$alkanediyl optionally substituted with one or more substituents individually selected from the group of substituents consisting of hydroxy, $C_{1-6}$alkyloxy, $Ar^1C_{1-6}$alkyloxy, $C_{1-6}$alkylthio, $Ar^1C_{1-6}$alkylthio, $HO(-CH_2-CH_2-O)_n-$, $C_{1-6}$alkyloxy$(-CH_2-CH_2-O)_n-$ or $Ar^1C_{1-6}$alkyloxy$(-CH_2-CH_2-O)_n-$;

$R^1$ is $Ar^1$ or a monocyclic or bicyclic heterocycle being selected from piperidinyl, piperazinyl, pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, furanyl, tetrahydrofuranyl, thienyl, pyrrolyl, thiazolyl, oxazolyl, imidazolyl, isothiazolyl, pyrazolyl, isoxazolyl, oxadiazolyl, quinolinyl, quinoxalinyl, benzofuranyl, benzothienyl, benzimidazolyl, benzoxazolyl, benzthiazolyl, pyridopyridyl, naphthiridinyl, 1*H*-imidazo[4,5-b]pyridinyl, 3*H*-imidazo[4,5-b]pyridinyl, imidazo[1,2-a]-pyridinyl, 2,3-dihydro-1,4-dioxino[2,3-b]pyridyl or a radical of formula



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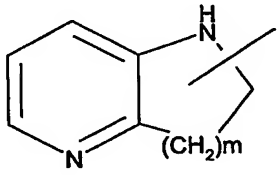

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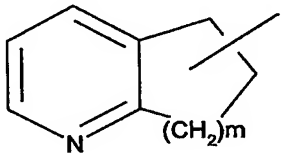

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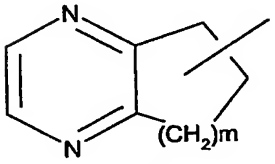

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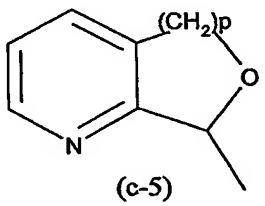

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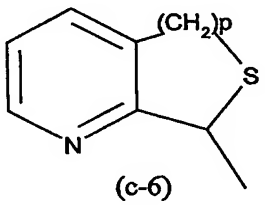

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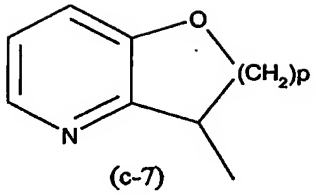


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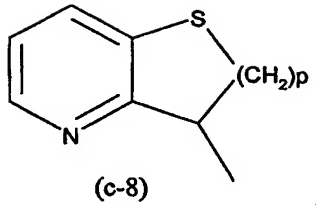


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;

wherein each of said monocyclic or bicyclic heterocycles may optionally be substituted with 1 or where possible more, such as 2, 3, 4 or 5, substituents individually selected from the group of substituents consisting of halo, hydroxy, amino, cyano, carboxyl, $C_{1-6}$alkyl, $C_{1-6}$alkyloxy, $C_{1-6}$alkylthio, $C_{1-6}$alkyloxy$C_{1-6}$alkyl, $Ar^1$, $Ar^1C_{1-6}$alkyl, $Ar^1C_{1-6}$alkyloxy, hydroxy$C_{1-6}$alkyl, mono-or di($C_{1-6}$alkyl)amino, mono-or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyl, $C_{1-6}$alkylcarbonylamino, $C_{1-6}$alkyl-$SO_2$-$NR^{5c}$-, $Ar^1$-$SO_2$-$NR^{5c}$-, $C_{1-6}$alkyloxycarbonyl, -C(=O)-$NR^{5c}R^{5d}$, HO(-$CH_2$-$CH_2$-O)$_n$-, halo(-$CH_2$-$CH_2$-O)$_n$-, $C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$-, $Ar^1C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$- and mono-or di($C_{1-6}$alkyl)amino(-$CH_2$-$CH_2$-O)$_n$-;

each n independently is 1, 2, 3 or 4;

each m independently is 1 or 2;

each p independently is 1 or 2;

each t independently is 0, 1 or 2;

Q is $R^7$, pyrrolidinyl substituted with $R^7$, piperidinyl substituted with $R^7$ or homo-piperidinyl substituted with $R^7$ wherein $R^7$ is $C_{1-6}$alkyl substituted with a heterocycle selected from the group consisting of morpholinyl, thiomorpholinyl, 1-oxo-thiomorpholinyl and 1,1-dioxothiomorpholinyl, wherein each of said heterocyle may be optionally substituted with one or two substituents selected from the group consisting of hydroxy, carboxyl, $C_{1-4}$alkyloxycarbonyl, aminocarbonyl, mono- or di($C_{1-4}$alkyl)-aminocarbonyl, $C_{1-4}$alkylcarbonylamino, aminosulfonyl and mono- or di($C_{1-4}$alkyl)-aminosulfonyl;

one of $R^{2a}$ and $R^{3a}$ is selected from halo, optionally mono- or polysubstituted $C_{1-6}$alkyl, optionally mono- or polysubstituted $C_{2-6}$alkenyl, nitro, hydroxy, $Ar^2$, $N(R^{4a}R^{4b})$, $N(R^{4a}R^{4b})$sulfonyl, $N(R^{4a}R^{4b})$carbonyl, $C_{1-6}$alkyloxy, $Ar^2$oxy, $Ar^2C_{1-6}$alkyloxy, carboxyl, $C_{1-6}$alkyloxycarbonyl, or -C(=Z)$Ar^2$; and the other one of $R^{2a}$ and $R^{3a}$ is hydrogen;

wherein

- =Z is =O, =CH-C(=O)-$NR^{5a}R^{5b}$, $=CH_2$, =CH-$C_{1-6}$alkyl, =N-OH or =N-O-$C_{1-6}$alkyl; and
- the optional substituents on $C_{1-6}$alkyl and $C_{2-6}$alkenyl can be the same or can be different relative to one another, and are each independently selected from the group of substituents consisting of hydroxy, cyano, halo, nitro, $N(R^{4a}R^{4b})$,

$N(R^{4a}R^{4b})$sulfonyl, Het, $Ar^2$, $C_{1-6}$alkyloxy, $C_{1-6}$alkyl-$S(=O)_t$, $Ar^2$oxy, $Ar^2$-$S(=O)_t$, $Ar^2C_{1-6}$alkyloxy, $Ar^2C_{1-6}$alkyl-$S(=O)_t$, Het-oxy, Het-$S(=O)_t$, $HetC_{1-6}$alkyloxy, $HetC_{1-6}$alkyl-$S(=O)_t$, carboxyl, $C_{1-6}$alkyloxycarbonyl and $-C(=Z)Ar^2$;

in case $R^{2a}$ is different from hydrogen then $R^{2b}$ is hydrogen, $C_{1-6}$alkyl or halogen and $R^{3b}$ is hydrogen;

in case $R^{3a}$ is different from hydrogen then $R^{3b}$ is hydrogen, $C_{1-6}$alkyl or halogen and $R^{2b}$ is hydrogen;

$R^{4a}$ and $R^{4b}$ can be the same or can be different relative to one another, and are each independently selected from the group of substituents consisting of hydrogen, $C_{1-6}$alkyl, $Ar^2$, $Ar^2C_{1-6}$alkyl, $C_{1-6}$alkylcarbonyl, $Ar^2$carbonyl, $Ar^2C_{1-6}$alkyl-carbonyl, $C_{1-6}$alkylsulfonyl, $Ar^2$sulfonyl, $Ar^2C_{1-6}$alkylsulfonyl, $C_{1-6}$alkyloxy-$C_{1-6}$alkyl, amino$C_{1-6}$alkyl, mono- or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, hydroxy-$C_{1-6}$alkyl, Het, Het-$C_{1-6}$alkyl, Het-carbonyl, Het-sulfonyl, $HetC_{1-6}$alkylsulfonyl and Het-$C_{1-6}$alkylcarbonyl;

$R^{5a}$ and $R^{5b}$ can be the same or can be different relative to one another, and are each independently hydrogen or $C_{1-6}$alkyl; or

$R^{5a}$ and $R^{5b}$ taken together may form a bivalent radical of formula $-(CH_2)_s-$ wherein s is 4 or 5;

$R^{5c}$ and $R^{5d}$ can be the same or can be different relative to one another, and are each independently hydrogen or $C_{1-6}$alkyl; or

$R^{5c}$ and $R^{5d}$ taken together may form a bivalent radical of formula $-(CH_2)_s-$ wherein s is 4 or 5;

$R^{6a}$ is hydrogen, $C_{1-6}$alkyl, $Ar^1$, $Ar^1C_{1-6}$alkyl, $C_{1-6}$alkylcarbonyl, $Ar^1$carbonyl, $Ar^1C_{1-6}$alkylcarbonyl, $C_{1-6}$alkylsulfonyl, $Ar^1$sulfonyl, $Ar^1C_{1-6}$alkylsulfonyl, $C_{1-6}$alkyloxy$C_{1-6}$alkyl, amino$C_{1-6}$alkyl, mono- or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, hydroxy$C_{1-6}$alkyl, Het, Het-$C_{1-6}$alkyl, Het-carbonyl, Het-sulfonyl, Het-$C_{1-6}$alkylcarbonyl;

$R^{6b}$ is hydrogen, $C_{1-6}$alkyl, $Ar^1$ or $Ar^1C_{1-6}$alkyl;

$R^{6c}$ is $C_{1-6}$alkyl, $Ar^1$ or $Ar^1C_{1-6}$alkyl;

$Ar^1$ is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, $C_{1-6}$alkyl, hydroxy$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyl, and $C_{1-6}$alkyloxy;

$Ar^2$ is phenyl, phenyl annealed with a $C_{5-7}$cycloalkyl, or phenyl substituted with 1 or more, such as 2, 3, 4 or 5, substituents selected from halo, cyano, $C_{1-6}$alkyl, cyano$C_{1-6}$alkyl, cyano$C_{2-6}$alkenyl, $Ar^1$, $R^{6b}$-O-, $R^{6b}$-S-, $N(R^{6a}R^{6b})$, polyhalo$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyloxy, polyhalo$C_{1-6}$alkylthio, $R^{6c}$-C(=O)-, $R^{6b}$-O-C(=O)-, $N(R^{6a}R^{6b})$-C(=O)-, $R^{6b}$-O-$C_{1-6}$alkyl, $R^{6b}$-S-$C_{1-6}$alkyl,

$R^{6c}$-S(=O)$_2$-$C_{1-6}$alkyl, N($R^{6a}R^{6b}$)-$C_{1-6}$alkyl, $R^{6c}$-C(=O)-$C_{1-6}$alkyl, $R^{6b}$-O-C(=O)-$C_{1-6}$alkyl, N($R^{6a}R^{6b}$)-C(=O)-$C_{1-6}$alkyl, $R^{6c}$-C(=O)-$NR^{6b}$-, $R^{6c}$-C(=O)-O-, $R^{6c}$-C(=O)-$NR^{6b}$-$C_{1-6}$alkyl, $R^{6c}$-C(=O)-O-$C_{1-6}$alkyl, N($R^{6a}R^{6b}$)-S(=O)$_2$-;

Het is a heterocycle being selected from tetrahydrofuranyl, tetrahydrothienyl, pyrrolidinyl, pyrrolidinonyl, furanyl, thienyl, pyrrolyl, thiazolyl, oxazolyl, imidazolyl, isothiazolyl, pyrazolyl, isoxazolyl, oxadiazolyl, thiadiazolyl, piperidinyl, homopiperidinyl, piperazinyl, morpholinyl, pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, tetrahydroquinolinyl, quinolinyl, isoquinolinyl, benzodioxanyl, benzodioxolyl, indolinyl, indolyl, each of said heterocycle may optionally be substituted with oxo, amino, $Ar^1$, $C_{1-4}$alkyl, amino$C_{1-4}$alkyl, $Ar^1C_{1-4}$alkyl, mono- or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, mono- or di($C_{1-6}$alkyl)amino.

The term prodrug as used throughout this text means the pharmacologically acceptable derivatives, e.g. esters and amides, such that the resulting biotransformation product of the derivative is the active drug as defined in the compounds of formula (I). The reference by Goodman and Gilman (The Pharmacological Basis of Therapeutics, 8th ed., McGraw-Hill, Int. Ed. 1992, "Biotransformation of Drugs", p. 13-15) describing prodrugs generally, is hereby incorporated. Prodrugs are characterized by a good aqueous solubility and bioavailability, and are readily metabolized into the active inhibitors *in vivo*.

As used herein $C_{1-3}$alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 3 carbon atoms such as methyl, ethyl, propyl, 1-methylethyl and the like; $C_{1-4}$alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as the group defined for $C_{1-3}$alkyl and butyl and the like; $C_{2-4}$alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 2 to 4 carbon atoms such as ethyl, propyl, 1-methylethyl, butyl and the like; $C_{1-6}$alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 6 carbon atoms such as the groups defined for $C_{1-4}$alkyl and pentyl, hexyl, 2-methylbutyl and the like; $C_{1-9}$alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 9 carbon atoms such as the groups defined for $C_{1-6}$alkyl and heptyl, octyl, nonyl, 2-methylhexyl, 2-methylheptyl and the like; $C_{1-10}$alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 10 carbon atoms such as the groups defined for $C_{1-9}$alkyl and decyl, 2-methylnonyl and the like.

$C_{3-7}$cycloalkyl is generic to cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl.

$C_{2-5}$alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 2 to 5 carbon atoms such as, for example, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl, 1,2-propanediyl, 2,3-butanediyl, 1,5-pentanediyl and the like, $C_{1-4}$alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as, for example, methylene, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl and the like; $C_{1-6}$alkanediyl is meant to include $C_{1-4}$alkanediyl and the higher homologues thereof having from 5 to 6 carbon atoms such as, for example, 1,5-pentanediyl, 1,6-hexanediyl and the like; $C_{1-10}$alkanediyl is meant to include $C_{1-6}$alkanediyl and the higher homologues thereof having from 7 to 10 carbon atoms such as, for example, 1,7-heptanediyl, 1,8-octanediyl, 1,9-nonanediyl, 1,10-decanediyl and the like.

As used herein before, the term (=O) forms a carbonyl moiety when attached to a carbon atom, a sulfoxide moiety when attached to a sulfur atom and a sulfonyl moiety when two of said terms are attached to a sulfur atom. The term (=N-OH) forms a hydroxylimine moiety when attached to a carbon atom.

The term halo is generic to fluoro, chloro, bromo and iodo. As used in the foregoing and hereinafter, polyhalo$C_{1-6}$alkyl as a group or part of a group is defined as mono- or polyhalosubstituted $C_{1-6}$alkyl, in particular methyl with one or more fluoro atoms, for example, difluoromethyl or trifluoromethyl. In case more than one halogen atoms are attached to an alkyl group within the definition of polyhalo$C_{1-4}$alkyl, they may be the same or different.

It should be noted that the radical positions on any molecular moiety used in the definitions may be anywhere on such moiety as long as it is chemically stable.

Radicals used in the definitions of the variables include all possible isomers unless otherwise indicated. For instance pyridyl includes 2-pyridyl, 3-pyridyl and 4-pyridyl; pentyl includes 1-pentyl, 2-pentyl and 3-pentyl.

When any variable occurs more than one time in any constituent, each definition is independent.

Whenever used hereinafter, the term "compounds of formula (I)", or "the present compounds" or similar term is meant to include the compounds of general formula (I), their prodrugs, *N*-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms. An interesting subgroup of the compounds of formula (I) or any subgroup thereof are the *N*-oxides, salts and all the stereoisomeric forms of the compounds of formula (I).

It will be appreciated that some of the compounds of formula (I) may contain one or more centers of chirality and exist as stereochemically isomeric forms.

The term "stereochemically isomeric forms" as used hereinbefore defines all the possible compounds made up of the same atoms bonded by the same sequence of bonds but having different three-dimensional structures which are not interchangeable, which the compounds of formula (I) may possess.

Unless otherwise mentioned or indicated, the chemical designation of a compound encompasses the mixture of all possible stereochemically isomeric forms which said compound may possess. Said mixture may contain all diastereomers and/or enantiomers of the basic molecular structure of said compound. All stereochemically isomeric forms of the compounds of the present invention both in pure form or in admixture with each other are intended to be embraced within the scope of the present invention.

Pure stereoisomeric forms of the compounds and intermediates as mentioned herein are defined as isomers substantially free of other enantiomeric or diastereomeric forms of the same basic molecular structure of said compounds or intermediates. In particular, the term 'stereoisomerically pure' concerns compounds or intermediates having a stereoisomeric excess of at least 80% (i. e. minimum 90% of one isomer and maximum 10% of the other possible isomers) up to a stereoisomeric excess of 100% (i.e. 100% of one isomer and none of the other), more in particular, compounds or intermediates having a stereoisomeric excess of 90% up to 100%, even more in particular having a stereoisomeric excess of 94% up to 100% and most in particular having a stereoisomeric excess of 97% up to 100%. The terms 'enantiomerically pure' and 'diastereomerically pure' should be understood in a similar way, but then having regard to the enantiomeric excess, respectively the diastereomeric excess of the mixture in question.

Pure stereoisomeric forms of the compounds and intermediates of this invention may be obtained by the application of art-known procedures. For instance, enantiomers may be separated from each other by the selective crystallization of their diastereomeric

salts with optically active acids or bases. Examples thereof are tartaric acid, dibenzoyltartaric acid, ditoluoyltartaric acid and camphosulfonic acid. Alternatively, enantiomers may be separated by chromatographic techniques using chiral stationary phases. Said pure stereochemically isomeric forms may also be derived from the corresponding pure stereochemically isomeric forms of the appropriate starting materials, provided that the reaction occurs stereospecifically. Preferably, if a specific stereoisomer is desired, said compound will be synthesized by stereospecific methods of preparation. These methods will advantageously employ enantiomerically pure starting materials.

The diastereomeric racemates of formula (I) can be obtained separately by conventional methods. Appropriate physical separation methods that may advantageously be employed are, for example, selective crystallization and chromatography, e.g. column chromatography.

For some of the compounds of formula (I), their prodrugs, N-oxides, salts, solvates, quaternary amines, or metal complexes and the intermediates used in the preparation thereof, the absolute stereochemical configuration was not experimentally determined. A person skilled in the art is able to determine the absolute configuration of such compounds using art-known methods such as, for example, X-ray diffraction.

The present invention is also intended to include all isotopes of atoms occurring on the present compounds. Isotopes include those atoms having the same atomic number but different mass numbers. By way of general example and without limitation, isotopes of hydrogen include tritium and deuterium. Isotopes of carbon include C-13 and C-14.

For therapeutic use, salts of the compounds of formula (I) are those wherein the counterion is pharmaceutically acceptable. However, salts of acids and bases which are non-pharmaceutically acceptable may also find use, for example, in the preparation or purification of a pharmaceutically acceptable compound. All salts, whether pharmaceutically acceptable or not are included within the ambit of the present invention.

The pharmaceutically acceptable acid and base addition salts as mentioned hereinabove are meant to comprise the therapeutically active non-toxic acid and base addition salt forms which the compounds of formula (I) are able to form. The pharmaceutically acceptable acid addition salts can conveniently be obtained by treating the base form with such appropriate acid. Appropriate acids comprise, for example, inorganic acids such as hydrohalic acids, e.g. hydrochloric or hydrobromic acid, sulfuric, nitric, phosphoric and the like acids; or organic acids such as, for example, acetic, propanoic, hydroxyacetic, lactic, pyruvic, oxalic (i.e. ethanedioic), malonic, succinic (i.e. butane-

dioic acid), maleic, fumaric, malic (i.e. hydroxybutanedioic acid), tartaric, citric, methanesulfonic, ethanesulfonic, benzenesulfonic, *p*-toluenesulfonic, cyclamic, salicylic, *p*-aminosalicylic, pamoic and the like acids.

Conversely said salt forms can be converted by treatment with an appropriate base into the free base form.

The compounds of formula (I) containing an acidic proton may also be converted into their non-toxic metal or amine addition salt forms by treatment with appropriate organic and inorganic bases. Appropriate base salt forms comprise, for example, the ammonium salts, the alkali and earth alkaline metal salts, e.g. the lithium, sodium, potassium, magnesium, calcium salts and the like, salts with organic bases, e.g. the benzathine, *N*-methyl-D-glucamine, hydrabamine salts, and salts with amino acids such as, for example, arginine, lysine and the like.

The term addition salt as used hereinabove also comprises the solvates which the compounds of formula (I) as well as the salts thereof, are able to form. Such solvates are for example hydrates, alcoholates and the like.

The term "quaternary amine" as used hereinbefore defines the quaternary ammonium salts which the compounds of formula (I) are able to form by reaction between a basic nitrogen of a compound of formula (I) and an appropriate quaternizing agent, such as, for example, an optionally substituted alkylhalide, arylhalide or arylalkylhalide, e.g. methyliodide or benzyliodide. Other reactants with good leaving groups may also be used, such as alkyl trifluoromethanesulfonates, alkyl methanesulfonates, and alkyl p-toluenesulfonates. A quaternary amine has a positively charged nitrogen. Pharmaceutically acceptable counterions include chloro, bromo, iodo, trifluoroacetate and acetate. The counterion of choice can be introduced using ion exchange resins.

The *N*-oxide forms of the present compounds are meant to comprise the compounds of formula (I) wherein one or several nitrogen atoms are oxidized to the so-called *N*-oxide.

It will be appreciated that the compounds of formula (I) may have metal binding, chelating, complexating properties and therefore may exist as metal complexes or metal chelates. Such metalated derivatives of the compounds of formula (I) are intended to be included within the scope of the present invention.

Some of the compounds of formula (I) may also exist in their tautomeric form. Such forms although not explicitly indicated in the above formula are intended to be included

within the scope of the present invention.

Interesting compounds are those compounds of formula (I) or any subgroup thereof wherein G is $C_{1-10}$alkanediyl; more in particular, G is methylene.

Other interesting compounds are those compounds of formula (I) or any subgroup thereof wherein $R^1$ is pyridyl optionally substituted with 1 or 2 substituents individually selected from the group of substituents consisting of halo, hydroxy, amino, cyano, carboxyl, $C_{1-6}$alkyl, $C_{1-6}$alkyloxy, $C_{1-6}$alkylthio, $C_{1-6}$alkyloxy$C_{1-6}$alkyl, $Ar^1$, $Ar^1C_{1-6}$alkyl, $Ar^1C_{1-6}$alkyloxy, hydroxy$C_{1-6}$alkyl, mono-or di($C_{1-6}$alkyl)amino, mono-or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyl, $C_{1-6}$alkylcarbonylamino, $C_{1-6}$alkyl-$SO_2$-$NR^{5c}$-, $Ar^1$-$SO_2$-$NR^{5c}$-, $C_{1-6}$alkyloxycarbonyl, -C(=O)-$NR^{5c}R^{5d}$, HO(-$CH_2$-$CH_2$-O)$_n$-, halo(-$CH_2$-$CH_2$-O)$_n$-, $C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$-, $Ar^1C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$- and mono-or di($C_{1-6}$alkyl)amino(-$CH_2$-$CH_2$-O)$_n$-; more in particular $R^1$ is pyridyl substituted with hydroxy and $C_{1-6}$alkyl.

Other interesting compounds are those compounds of formula (I) wherein t is 2.

Preferred compounds are those compounds listed in tables 1 through 8, more in particular the compound numbers 1 to 24, 26 to 77, 80 to 83, 90 to 92, 97 and 99.

Compounds of formula (I) may be converted into each other following art-known functional group transformation reactions, comprising those described hereinafter.

The compounds of formula (I) may be converted to the corresponding *N*-oxide forms following art-known procedures for converting a trivalent nitrogen into its *N*-oxide form. Said *N*-oxidation reaction may generally be carried out by reacting the starting material of formula (I) with an appropriate organic or inorganic peroxide. Appropriate inorganic peroxides comprise, for example, hydrogen peroxide, alkali metal or earth alkaline metal peroxides, e.g. sodium peroxide, potassium peroxide; appropriate organic peroxides may comprise peroxy acids such as, for example, benzenecarboperoxoic acid or halo substituted benzenecarboperoxoic acid, e.g. 3-chlorobenzenecarboperoxoic acid, peroxoalkanoic acids, e.g. peroxoacetic acid, alkylhydroperoxides, e.g. t.butyl hydro-peroxide. Suitable solvents are, for example, water, lower alcohols, e.g. ethanol and the like, hydrocarbons, e.g. toluene, ketones, e.g. 2-butanone, halogenated hydrocarbons, e.g. dichloromethane, and mixtures of such solvents.

Pure stereochemically isomeric forms of the compounds of formula (I) may be obtained by the application of art-known procedures. Diastereomers may be separated by physical

methods such as selective crystallization and chromatographic techniques, e.g., counter-current distribution, liquid chromatography and the like.

The compounds of formula (I) as prepared in the hereinabove described processes are generally racemic mixtures of enantiomers which can be separated from one another following art-known resolution procedures. The racemic compounds of formula (I) which are sufficiently basic or acidic may be converted into the corresponding diastereomeric salt forms by reaction with a suitable chiral acid, respectively chiral base. Said diastereomeric salt forms are subsequently separated, for example, by selective or fractional crystallization and the enantiomers are liberated therefrom by alkali or acid. An alternative manner of separating the enantiomeric forms of the compounds of formula (I) involves liquid chromatography, in particular liquid chromatography using a chiral stationary phase. Said pure stereochemically isomeric forms may also be derived from the corresponding pure stereochemically isomeric forms of the appropriate starting materials, provided that the reaction occurs stereospecifically. Preferably if a specific stereoisomer is desired, said compound will be synthesized by stereospecific methods of preparation. These methods will advantageously employ enantiomerically pure starting materials.

The compounds of formula (I) show antiviral properties. Viral infections treatable using the compounds and methods of the present invention include those infections brought on by ortho- and paramyxoviruses and in particular by human and bovine respiratory syncytial virus (RSV).

The *in vitro* antiviral activity against RSV of the present compounds was tested in a test as described in the experimental part of the description, and may also be demonstrated in a virus yield reduction assay. The *in vivo* antiviral activity against RSV of the present compounds may be demonstrated in a test model using cotton rats as described in Wyde et al. (Antiviral Research (1998), 38, 31-42).

Due to their antiviral properties, particularly their anti-RSV properties, the compounds of formula (I) or any subgroup thereof, their prodrugs, *N*-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms, are useful in the treatment of individuals experiencing a viral infection, particularly a RSV infection, and for the prophylaxis of these infections. In general, the compounds of the present invention may be useful in the treatment of warm-blooded animals infected with viruses, in particular the respiratory syncytial virus.

The compounds of the present invention or any subgroup thereof may therefore be used as medicines. Said use as a medicine or method of treatment comprises the systemic

administration to viral infected subjects or to subjects susceptible to viral infections of an amount effective to combat the conditions associated with the viral infection, in particular the RSV infection.

The present invention also relates to the use of the present compounds or any subgroup thereof in the manufacture of a medicament for the treatment or the prevention of viral infections, particularly RSV infection.

The compounds of the present invention or any subgroup thereof may be formulated into various pharmaceutical forms for administration purposes. As appropriate compositions there may be cited all compositions usually employed for systemically administering drugs. To prepare the pharmaceutical compositions of this invention, an effective amount of the particular compound, optionally in addition salt form or metal complex, as the active ingredient is combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirable in unitary dosage form suitable, particularly, for administration orally, rectally, percutaneously, or by parenteral injection. For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs, emulsions and solutions; or solid carriers such as starches, sugars, kaolin, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules, and tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit forms, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example, to aid solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. Also included are solid form preparations which are intended to be converted, shortly before use, to liquid form preparations. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined with suitable additives of any nature in minor proportions, which additives do not introduce a significant deleterious effect on the skin.

The compounds of the present invention may also be administered via oral inhalation or insufflation by means of methods and formulations employed in the art for administration via this way. Thus, in general the compounds of the present invention may be administered to the lungs in the form of a solution, a suspension or a dry powder, a solution being preferred. Any system developed for the delivery of solutions, suspensions or dry powders via oral inhalation or insufflation are suitable for the administration of the present compounds.
Thus, the present invention also provides a pharmaceutical composition adapted for administration by inhalation or insufflation through the mouth comprising a compound of formula (I) and a pharmaceutically acceptable carrier. Preferably, the compounds of the present invention are administered via inhalation of a solution in nebulized or aerosolized doses.

It is especially advantageous to formulate the aforementioned pharmaceutical compositions in unit dosage form for ease of administration and uniformity of dosage. Unit dosage form as used herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such unit dosage forms are tablets (including scored or coated tablets), capsules, pills, suppositories, powder packets, wafers, injectable solutions or suspensions and the like, and segregated multiples thereof.

In general it is contemplated that an antivirally effective daily amount would be from 0.01 mg/kg to 500 mg/kg body weight, more preferably from 0.1 mg/kg to 50 mg/kg body weight. It may be appropriate to administer the required dose as two, three, four or more sub-doses at appropriate intervals throughout the day. Said sub-doses may be formulated as unit dosage forms, for example, containing 1 to 1000 mg, and in particular 5 to 200 mg of active ingredient per unit dosage form.

The exact dosage and frequency of administration depends on the particular compound of formula (I) used, the particular condition being treated, the severity of the condition being treated, the age, weight, sex, extent of disorder and general physical condition of the particular patient as well as other medication the individual may be taking, as is well known to those skilled in the art. Furthermore, it is evident that said effective daily amount may be lowered or increased depending on the response of the treated subject and/or depending on the evaluation of the physician prescribing the compounds of the instant invention. The effective daily amount ranges mentioned hereinabove are therefore only guidelines.

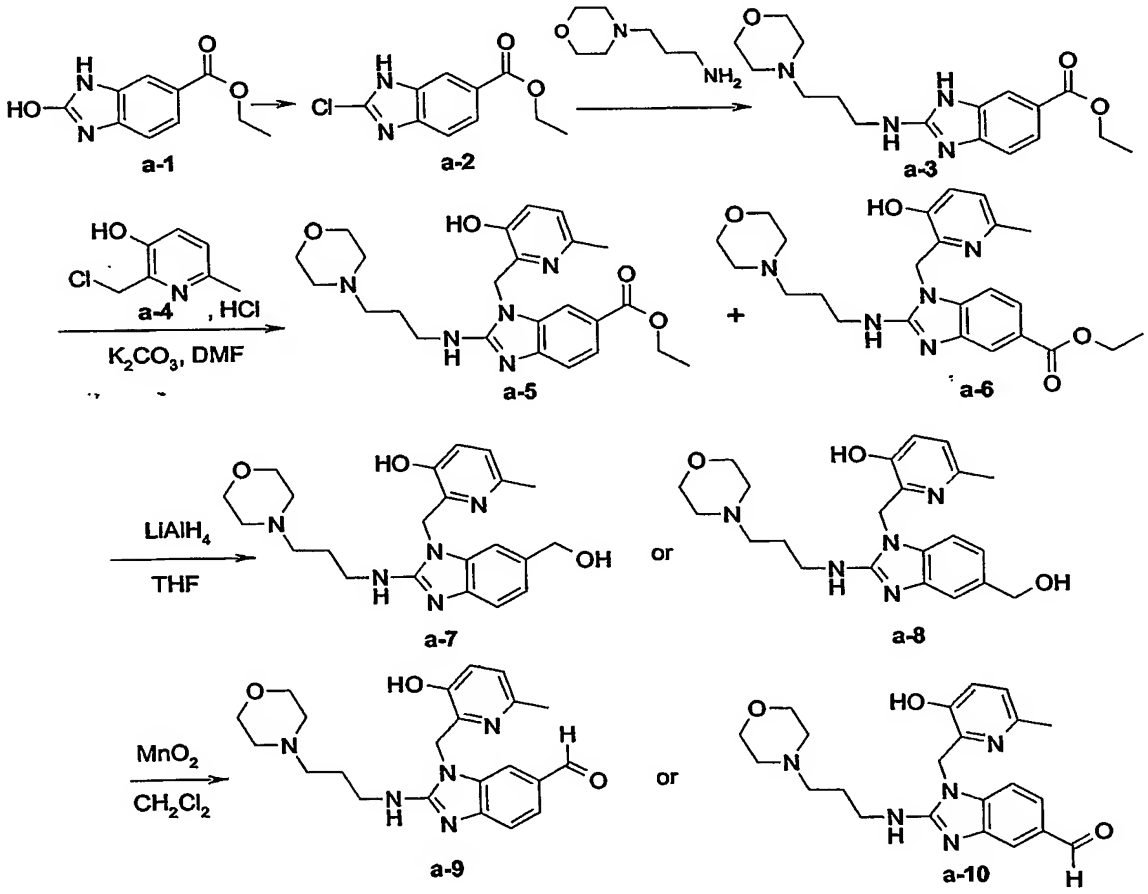
Also, the combination of another antiviral agent and a compound of formula (I) can be used as a medicine. Thus, the present invention also relates to a product containing (a) a compound of formula (I), and (b) another antiviral compound, as a combined preparation for simultaneous, separate or sequential use in antiviral treatment. The different drugs may be combined in a single preparation together with pharmaceutically acceptable carriers. For instance, the compounds of the present invention may be combined with interferon-beta or tumor necrosis factor-alpha in order to treat or prevent RSV infections.

**Experimental Part**

The following examples are intended to illustrate the present invention.

***A. Chemical synthesis of the compounds of formula (I)***

Scheme A-1




A mixture of 3,4-diamino benzoic acid ethyl ester (0.166 mol) and urea (0.199 mol) in xylene (300 ml) was stirred under reflux for 12 hours. The reaction was cooled down to

room temperature. The precipitate was filtered off, rinsed with xylene and diisopropylether, and then dried, yielding 32g of intermediate **a-1** (93%, melting point: > 260°C).

A mixture of **a-1** (0.073 mol) in $POCl_3$ (150 ml) was stirred at 100°C. HCl conc. (around 1.5 ml) was added drop wise very carefully until the dissolution of **a-1**. The mixture was stirred at 120°C for 6 hours. The solvent was evaporated till dryness. The residue was taken-up in $H_2O$/ice, basified with $K_2CO_3$ (powder) and extracted with ethylacetate + 10% methanol. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness, yielding 13.5 g of intermediate **a-2** (83%, melting point: 178°C).

A mixture of **a-2** (0.0356 mol) and N-propylamino-morpholine (0.0427 mol) was stirred at 120°C for 4 hours, and then taken up in $CH_2Cl_2$/$CH_3OH$. The organic layer was washed with a 10% solution of $K_2CO_3$ in water, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (11.9g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2$/$CH_3OH$/$NH_4OH$ 94/6/0.2; 15-40µm). The pure fractions were collected and the solvent was evaporated, yielding 6g of intermediate **a-3** (47%).

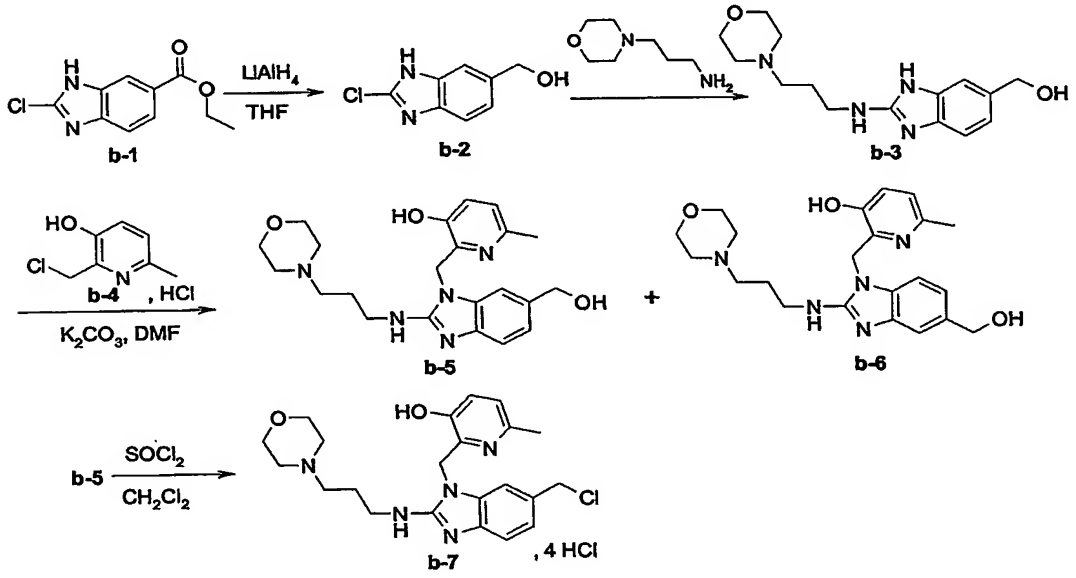
A mixture of **a-3** (0.018 mol), **a-4** (0.027 mol) and $K_2CO_3$ (0.054 mol) in $CH_3CN$ (100ml) and dimethylformamide (10ml) was stirred at 80°C for 12 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2$/$H_2O$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was crystallized from 2-propanone. The precipitate was filtered, washed with $H_2O$ and dried, yielding 2.8g of intermediate **a-6** (34%, melting point: 176°C). The mother layer was evaporated till dryness and purified by chromatography over silica gel (eluent: $CH_2Cl_2$/$CH_3OH$/$NH_4OH$ 93/7/0.7; 15-40µm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from $CH_3CN$/diisopropylether, yielding 1.6g of intermediate **a-5** (20%, melting point: 184°C).

A mixture of **a-5** (0.0035 mol) in tetrahydrofuran (60ml) was cooled down to 5°C under $N_2$ flow. $LiAlH_4$ (0.0105 mol) was added portion wise. The mixture was stirred at 5°C for 1 hour, and then stirred at room temperature for 2 hours. A minimum of $H_2O$ was added. $CH_2Cl_2$ was added. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was crystallized from 2-propanone/diisopropylether. The precipitate was filtered off and dried, yielding 1.2g of intermediate **a-7** (83%). Part of this fraction (0.1g) was crystallized from

2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.074g (melting point: 192°C). Intermediate **a-8** (melting point: 134°C) was prepared in an analogous way.

A mixture of **a-7** (0.0024 mol) and $MnO_2$ (2g) in $CH_2Cl_2$ (50ml) was stirred at room temperature for 12 hours, and then filtered over celite. Celite was washed with $H_2O$. The solvent of the filtrate was evaporated till dryness, yielding 0.9g of intermediate **a-9** (90%, melting point: 206°C). Intermediate **a-10** was prepared in an analogous way.

Scheme B



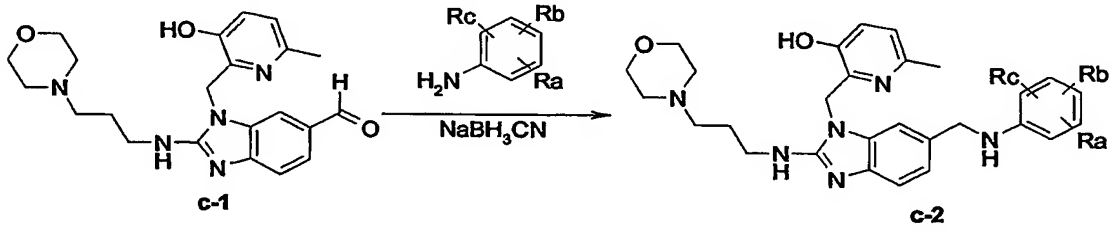

$LiAlH_4$ (0.146 mol) was added portion wise to a solution of tetrahydrofuran (200 ml) at 5°C under $N_2$ flow. A solution of **b-1** (0.073 mol) in tetrahydrofuran (200 ml) was then added drop wise. The mixture was stirred at 5°C for 3 hours. A minimum of $H_2O$ was then added, followed by a solution of $CH_2Cl_2/CH_3OH$ (90/10). The resulting mixture was dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness, yielding 12.6g of intermediate **b-2** (95%, melting point: 179°C).

A mixture of **b-2** (0.069 mol) and N-propylamino-morpholine (0.207 mol) was stirred at 125°C for 4 hours, and then taken up in $CH_2Cl_2/CH_3OH$. The organic layer was washed with a 10% solution of $K_2CO_3$ in water, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (37g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/0.5; 20-45μm). The pure fractions were collected and the solvent was evaporated, yielding 16.5g of intermediate **b-3** (82%).

A mixture of **b-3** (0.0396 mol), **b-4** (0.0475 mol) and $K_2CO_3$ (0.1188 mol) in dimethyl-formamide (110ml) was stirred at room temperature for 12 hours. The reaction was poured out into ice/water. The aqueous layer was saturated with $K_2CO_3$ (powder) and extracted with a solution of $CH_2Cl_2/CH_3OH$ (95/5). The residue was purified by chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/1; 20-45µm). The pure fractions were collected and the solvent was evaporated, yielding 5.4g of intermediate **b-5** (33%, melting point: 192°C) and 5g of intermediate **b-6** (31%, melting point: 134°C).

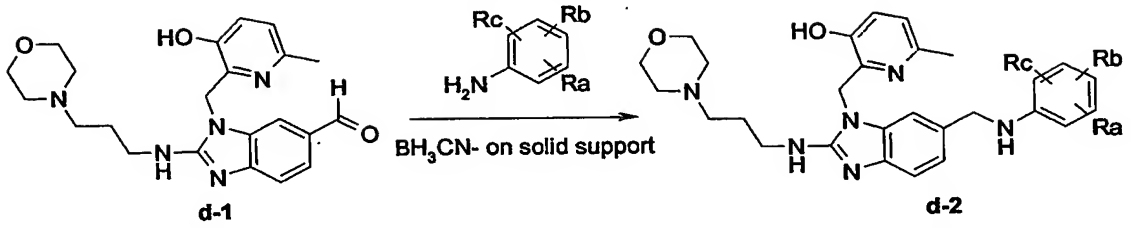
$SOCl_2$ (0.81ml) was added drop wise to a mixture of **b-5** (0.0006 mol) in $CH_2Cl_2$ (10ml) at 5°C. The mixture was stirred at 5°C for 2 hours, then brought to room temperature and stirred for 12 hours. The solvent was evaporated till dryness, yielding 0.42g of intermediate **b-7** (100%).

Scheme C



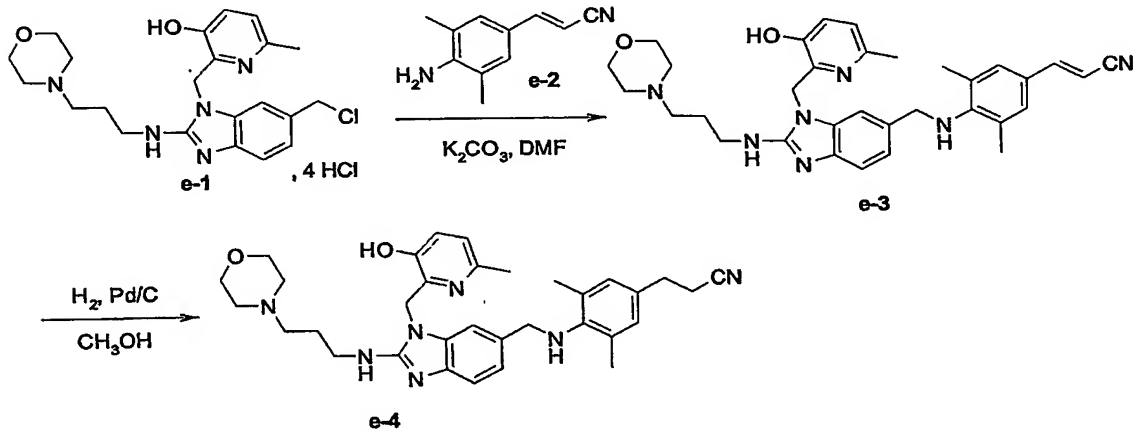

$CH_3CO_2H$ (0.2ml) was added at room temperature to a mixture of **c-1** (0.0004 mol), 3,5-dimethyl-aniline (0.0005 mol) and $NaBH_3CN$ (0.0005 mol) in $CH_3CN$ (25ml). The mixture was stirred at room temperature for 30 minutes. $CH_3CO_2H$ (0.2ml) was added. The mixture was stirred at room temperature for 12 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2$. The organic layer was washed with a 10% solution of $K_2CO_3$ in water, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (0.24g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/0.2; 10µm). The pure fractions were collected and the solvent was evaporated. The residue (0.15g, 60%) was crystallized from 2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.121g of 2-[6-[(3,5-dimethyl-phenylamino)-methyl]-2-(3-morpholin-4-yl-propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (48%, melting point: 199°C).

Scheme D




$CH_3CO_2H$ (0.15ml) was added at room temperature to a mixture of **d-1** (0.00037 mol), 2-(2-amino-4-methyl-phenyl)-ethanol (0.00044 mol) and $BH_3CN$ on solid support (0.00055 mol) in $CH_3OH$ (10ml). The mixture was stirred at room temperature for 12 hours. The solid support was filtered off, rinsed with $CH_3OH$ and the filtrate was concentrated. The residue was taken up in a 10% solution of $K_2CO_3$ in water and extracted with $CH_2Cl_2/CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (0.22g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 93/7/0.5; 10µm). The pure fractions were collected and the solvent was evaporated. The residue (0.13g, 65%) was crystallized from 2-propanone/diisopropylether. The precipitate was filtered off and dried, yielding 0.114g of 2-[6-{[2-(2-hydroxy-ethyl)-5-methyl-phenylamino]-methyl}-2-(3-morpholin-4-yl-propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (57%, melting point: 199°C).

Scheme E



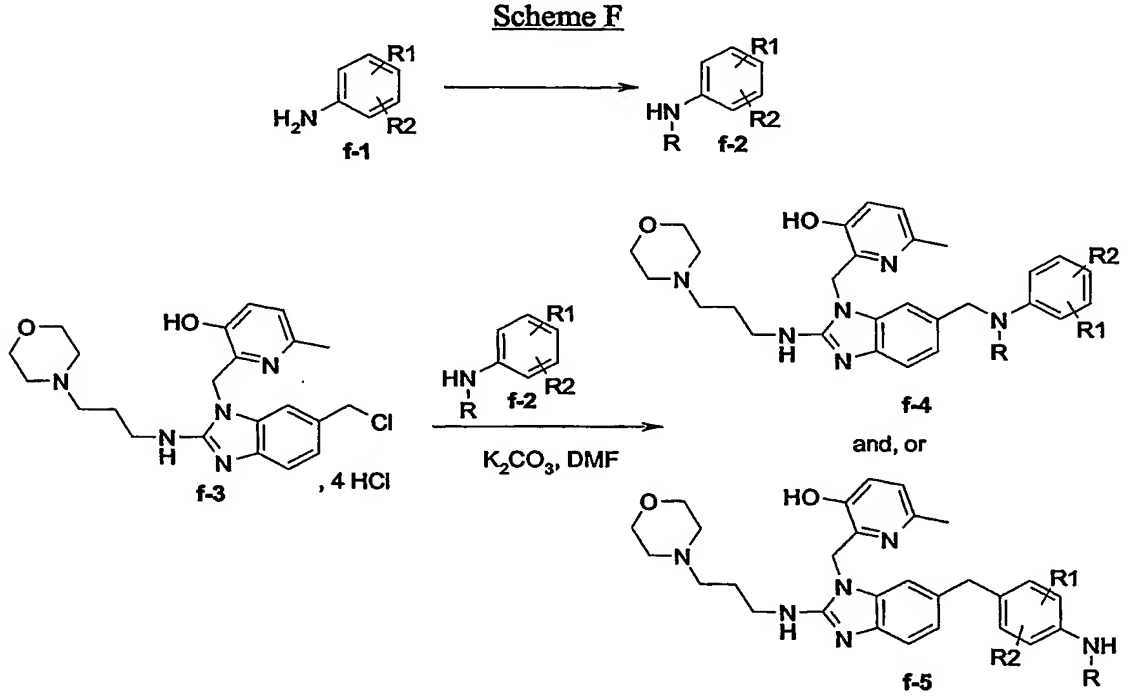

A mixture of **e-1** (0.000347 mol), **e-2** (0.00041 mol) and $K_2CO_3$ (0.00173 mol) in dimethylformamide (10 ml) was stirred at 80°C for 3 hours. The reaction was cooled down to room temperature and was poured out into a 10% solution of $K_2CO_3$ in water. The solution was saturated with $K_2CO_3$ (powder) and extracted with $CH_2Cl_2/ CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (0.15g) was purified by column chromato-

graphy over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 95/5/0.5; 10µm). The pure fractions were collected and the solvent was evaporated, yielding 0.03g of intermediate **e-3** (15%, mixture E/Z (89/11)).

A mixture of **e-3** (0.000106 mol) and Pd/C 10% (0.020g) in $CH_3OH$ (15 ml) and tetrahydrofuran (15 ml) was hydrogenated at room temperature for 6 hours under a 3 bar pressure. The reaction was filtered over celite. Celite was rinsed and the filtrate was evaporated till dryness. The residue (0.06g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 93/7/0.5; 10µm). The pure fractions were collected and the solvent was evaporated. The residue (0.028g) was crystallized from 2-propanone/diisopropylether, yielding 0.021g of 3-(4-{[3-(3-hydroxy-6-methyl-pyridin-2-ylmethyl)-2-(3-morpholin-4-yl-propylamino)-3H-benzoimidazol-5-yl-methyl]-amino}-3,5-dimethyl-phenyl)-propionitrile (35%, melting point: 114°C).

The isomers substituted in 5 position on the benzimidazole moiety have been synthesized analogous to the procedures described in schemes C and D, starting from intermediate **a-10**.

Scheme F




a. A mixture of 3-bromo-aniline (0.037 mol), 2-bromo-ethanol (0.074 mol) and triethylamine (0.0555 mol) in toluene (35ml) was stirred under reflux for 12 hours. The reaction was cooled down to room temperature and the precipitate was filtered off. The solvent of the filtrate was evaporated till dryness. The residue (22g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2$/

$CH_3OH/NH_4OH$ 98/2/0.1; 20-45μm). The pure fractions were collected and the solvent was evaporated, yielding 4.8g of 2-(3-bromo-phenylamino)-ethanol (60%). 5-(3,5-dimethyl-phenylamino)-pentanoic acid ethyl ester and 3-(3-bromo-phenylmino)-propionic acid ethyl ester and 4-*m*-tolylamino-butane-1-sulfonic acid amide and phosphoric acid 2-(3,5-dimethyl-phenylamino)-ethyl ester diethyl ester and [2-(3,5-dimethyl-phenylamino)-ethyl]-phosphonic acid diethyl ester and 4-*m*-tolylamino-butane-1-sulfonic acid methylamide were prepared analogously.

b. A mixture of 3,5-dimethyl-aniline (0.04 mol), 2-bromo-ethanol (0.033 mol) and $K_2CO_3$ (0.033 mol) in $CH_3CN$ (50 ml) was stirred at 80°C for 12 hours. The reaction was cooled down to room temperature and the solvent was evaporated. The residue was taken up in $CH_2Cl_2/CH_3OH$ (95/5) and washed with a saturated solution of $K_2CO_3$ in water. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 98/2/0.1; 20-45μm). The pure fractions were collected and the solvent was evaporated, yielding 1.9g of 2-(3,5-dimethyl-phenylamino)-ethanol (29%). 3-(3,5-dimethyl-phenylamino)-propionic acid ethyl ester and 4-(3,5-dimethyl-phenylamino)-butyric acid ethyl ester and (3,5-dimethyl-phenyl)-(2-morpholin-4-yl-ethyl)-amine and [2-(3,5-dimethyl-phenylamino)-ethyl]-carbamic acid tert-butyl ester were prepared analogously.

c. 3-(3,5-dimethyl-phenylamino)-propionic acid ethyl ester (0.0026 mol) in a 7N solution of $NH_3$ in $CH_3OH$ was stirred at 80°C in a sealed vessel. The reaction was cooled down to room temperature and the solvent was evaporated till dryness, yielding 0.5g of 3-(3,5-dimethyl-phenylamino)-propionamide (100%). 4-(3,5-dimethyl-phenylamino-butyramide and 4-*m*-tolylamino-butyramide and 3-*m*-tolylamino-propionamide and 3-(3-bromo-phenylamino)-propionamide were prepared analogously.

d. 3-(3,5-dimethyl-phenylamino)-propionic acid ethyl ester (0.00226 mol) in tetrahydrofuran (5 ml) was added drop wise to a slurry of $LiAlH_4$ (0.0034 mol) in tetrahydrofuran (10 ml) at 5°C under $N_2$ flow. The mixture was stirred at 5°C for 1 hour. A minimum of water and $CH_2Cl_2/CH_3OH$ (95/5) were added. The solution was dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness, yielding 0.35g of 3-(3,5-dimethyl-phenylamino-propan-1-ol (86%). 5-(3,5-dimethyl-phenylamino)-pentan-1-ol was prepared analogously.

A mixture of f-3 (0.000695 mol), 2-(3,5-dimethyl-phenylamino)-ethanol (0.0009 mol) and $K_2CO_3$ (0.0035 mol) in dimethylformamide (40ml) was stirred at 80°C for 4 hours. $H_2O$ was added. The solution was saturated with $K_2CO_3$ (powder) and extracted with

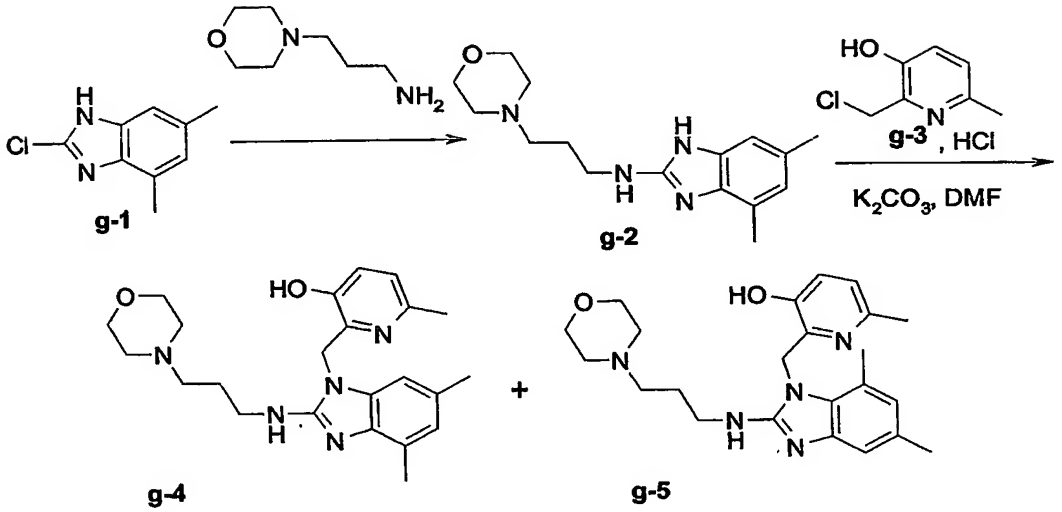
$CH_2Cl_2/CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated. The residue (0.5g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 93/7/0.5; 15-40µm). The pure fractions were collected and the solvent was evaporated, yielding 0.120g of fraction 1 (31%) and 0.045g of fraction 2 (12%). Fraction 1 was crystallized from $CH_3CN$/ diisopropylether. The precipitate was filtered, rinsed with diisopropylether and dried, yielding 0.1g of 2-[6-{[(3,5-dimethyl-phenyl)-(2-hydroxy-ethyl)-amino]-methyl}-2-(3-morpholin-4-yl-propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (example of compound **f-4**; 26%, melting point: 180°C). Fraction 2 was crystallized from 2-propanone/diisopropylether. The precipitate was filtered, rinsed with diisopropylether and dried, yielding 0.016g of 2-[6-[4-(2-hydroxy-ethylamino)-2,6-dimethyl-benzyl]-2-(3-morpholin-4-yl-propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (4%, melting point: 162°C).

A mixture of 4-{(3,5-dimethyl-phenyl)-[3-(3-hydroxy-6-methyl-pyridin-2-ylmethyl)-2-(3-morpholin-4-yl-propylamino)-3H-benzoimidazol-5-ylmethyl]-amino}-butyric acid ethyl ester, prepared as described for compounds **f-4**, (0.000175 mol) and $LiOH/H_2O$ (0.00035) in tetrahydrofuran (8 ml) and $H_2O$ (8 ml) was stirred at room temperature for 12 hours. The tetrahydrofuran was evaporated and a 1 N solution of NaOH in water was added. The solution was extracted with $CH_2Cl_2/CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated. The residue was taken up in $H_2O$. The precipitate was filtered off and dried, yielding 0.059g of 4-{(3,5-dimethyl-phenyl)-[3-(3-hydroxy-6-methyl-pyridin-2-ylmethyl)-2-(3-morpholin-4-yl-propylamino)-3H-benzoimidazol-5-ylmethyl]-amino}-butyric acid (56%, melting point: 121°C).

A mixture of (2-{(3,5-dimethyl-phenyl)-[3-(3-hydroxy-6-methyl-pyridin-2-ylmethyl)-2-(3-morpholin-4-yl-propylamino)-3H-benzoimidazol-5-ylmethyl]-amino}-ethyl)-carbamic acid tert-butyl ester, prepared as described for compounds **f-4**, (0.00012) in a 3N solution of HCl in water (10 ml) and tetrahydrofuran (10 ml) was stirred at room temperature for 12 hours. The precipitate was filtered off and taken up in a 10% solution of $K_2CO_3$ in water. The solution was saturated with $K_2CO_3$ (powder) and extracted with $CH_2Cl_2/CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (0.07g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 92/8/1; 10µm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from $CH_3CN/CH_3OH$/ diisopropylether, yielding 0.03g of 2-[6-{[(2-amino-ethyl)-(3,5-dimethyl-phenyl)-amino]-methyl}-2-(3-morpholin-4-yl-

propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (44%, melting point: 196°C).

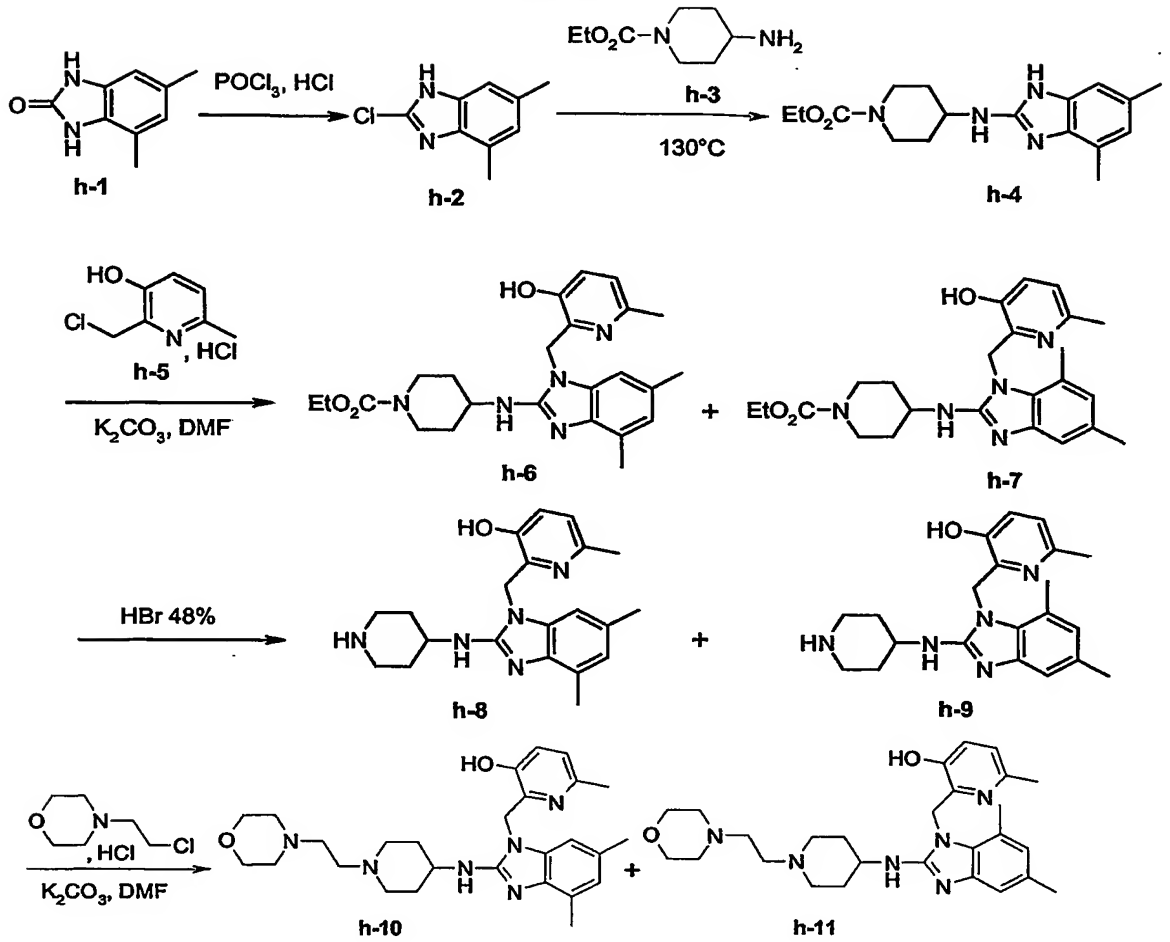
Scheme G




A mixture of **g-1** (0.0011 mol) and N-(propylamino)-morpholine (0.0044 mol) was stirred at 130°C for 4 hours, then brought to room temperature, taken up in $H_2O$ and extracted with $CH_2Cl_2$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated. The residue (0.328g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH$/triethylamine 99/1/0.1 to 90/10/1; 10µm). The pure fractions were collected and the solvent was evaporated, yielding 0.216g of intermediate **g-2** (68%).

A mixture of **g-2** (0.0007 mol), **g-3** (0.0008 mol) and $K_2CO_3$ (0.003 mol) in dimethylformamide (6ml) was stirred at 70°C for 12 hours, then brought to room temperature, taken up in $H_2O$ and extracted with $CH_2Cl_2$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated. The residue (0.5g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 93/7/0.5 then toluene/iPrOH/$NH_4OH$ 80/20/1; 10µm). Two fractions were collected and the solvent was evaporated, yielding 0.13g of fraction 1 and 0.036g of fraction 2. Fraction 1 was taken up in diisopropylether. The precipitate was filtered off and dried, yielding 0.1g of 2-[4,6-dimethyl-2-(3-morpholin-4-yl-propylamino)-benzoimidazol-1-yl-methyl]-6-methyl-pyridin-3-ol (33%, melting point: 228°C). Fraction 2 was taken up in diisopropylether. The precipitate was filtered off and dried, yielding 0.03g of 2-[5,7-dimethyl-2-(3-morpholin-4-yl-propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (10%, melting point: 234°C).

Scheme H




The mixture of **h-1** (0.06 mol) and $POCl_3$ (100 ml) was heated at 100°C and HCl 12N (2.5 ml) was added drop wise very carefully. The reaction was then stirred during 12 hours at 120°C and allowed to cool down to room temperature. The solvent was evaporated under reduced pressure and a 10% solution of potassium carbonate in water was added to the residue. The resulting precipitate was filtered off, rinsed with water and dried, yielding 10 g of **h-2** (93%, melting point = 152°C).

**h-2** (0.022 mol) and **h-3** (0.088 mol) were stirred at 130°C during 12 hours. The reaction was then allowed to cool down to room temperature, the residue was taken up in acetone and the precipitate was filtered off. The acetone solution was concentrated under reduced pressure. The residue was purified by column chromatography over silica gel (eluent: $CH_2Cl_2$/MeOH/$NH_4OH$ 95/5/0.1). The pure fractions were collected and the solvent was evaporated, yielding 5 g of **h-4** (72%).

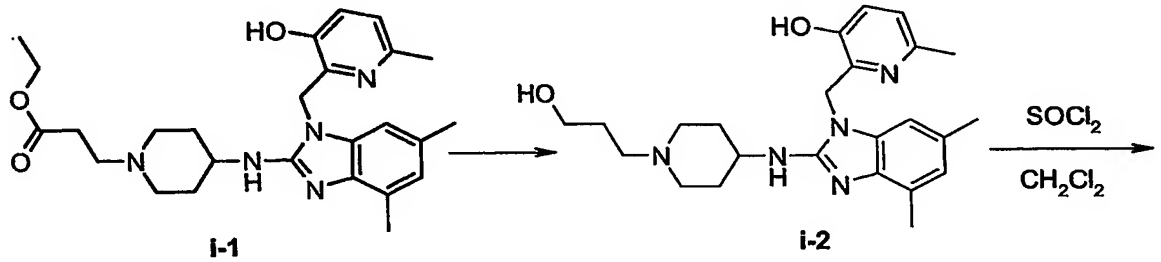
A mixture of **h-4** (0.0158 mol), **h-5** (0.019 mol) and potassium carbonate (0.0553 mol) in dimethylformamide (100ml) was stirred at 70°C for 24 hours. The solvent was

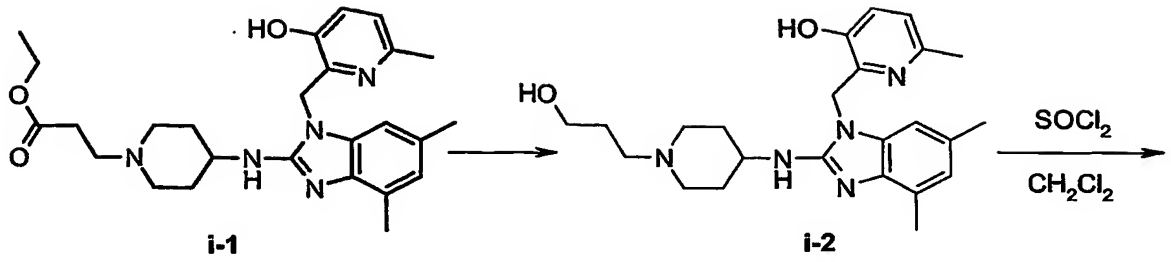
evaporated till dryness. The residue was taken up in $CH_2Cl_2/CH_3OH$ (90/10). The organic layer was washed with a 10% solution of $K_2CO_3$ in water, dried (over $MgSO_4$), filtered and the solvent was evaporated under reduced pressure. The residue was taken up in 2-propanone. The precipitate was filtered off, washed with $H_2O$ and dried, yielding 5g of **h-6 and h-7** (50/50 mixture, 73%).

A mixture of **h-6 and h-7** (0.0103 mol) in a 48% solution of HBr in water (50ml) was stirred at 60°C during 12 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2/CH_3OH$ (90/10). 10% solution of $K_2CO_3$ in water was added. The aqueous layer was saturated with $K_2CO_3$ (powder). The organic layer was separated, dried (over $MgSO_4$), filtered, and the solvent was evaporated till dryness, yielding 3.7g of **h-8 and h-9** (100%). This product was used directly in the next reaction step.

A mixture of **h-8** (0.0006 mol), **h-9** (0.0006 mol), N-(2-chloro-ethyl)-morpholine, HCl (0.0016 mol) and $K_2CO_3$ (0.0048 mol) in dimethylformamide (30ml) was stirred at room temperature for 48 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2$. The mixture was filtered. The filtrate was evaporated till dryness. The residue (1.2g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/0.5; 10µm). Two fractions were collected and the solvent was evaporated, yielding 0.023g of fraction 1 (4%) and 0.12g of fraction 2 (18%). Fraction 1 was crystallized from $CH_3OH/CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.02g of 2-[5,7-dimethyl-2-(2-morpholin-4-ylethyl-piperidin-4-ylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (3%, melting point: 226°C). Fraction 2 was crystallized from $CH_3OH$/ $CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.1g of 2-[4,6-dimethyl-2-(2-morpholin-4-ylethyl-piperidin-4-ylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (15%, melting point: 237°C).

Scheme I





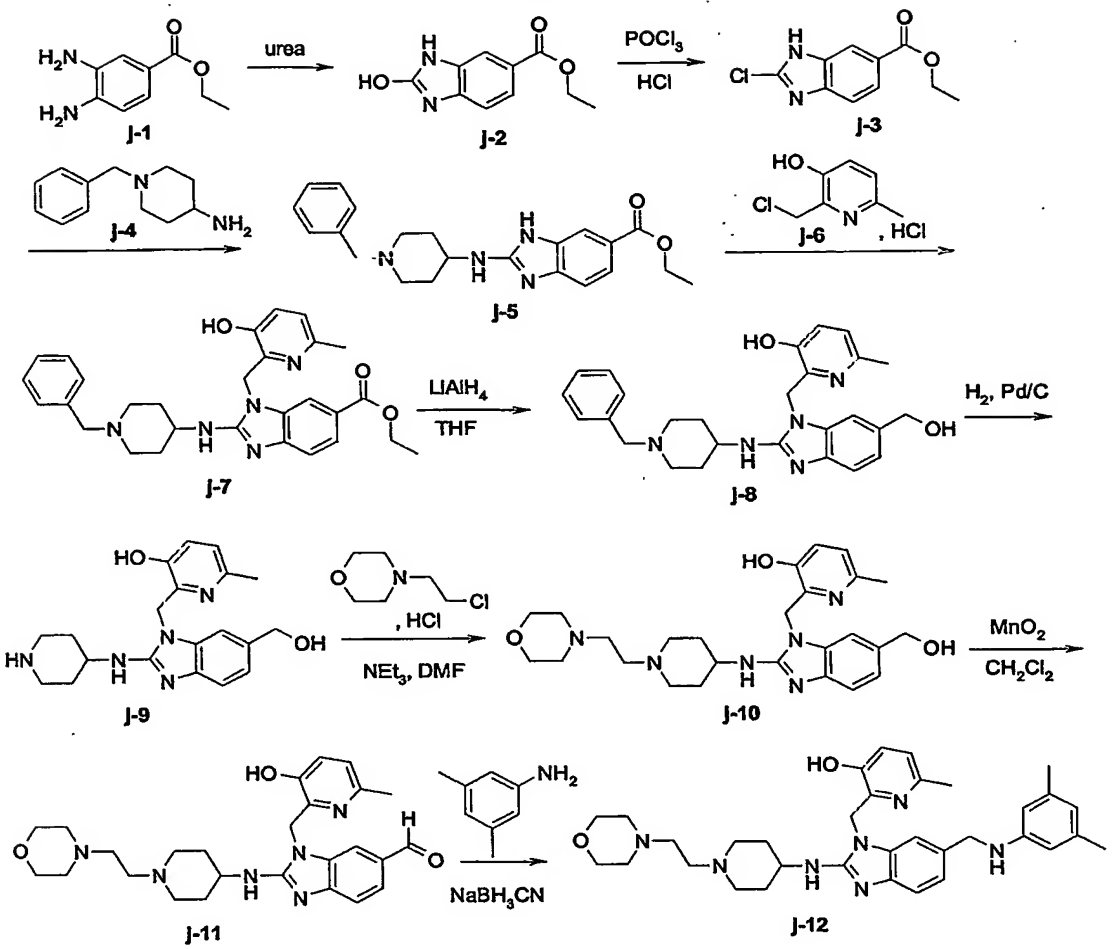

i-1 i-2

$LiAlH_4$ (0.0002 mol) was added at 5°C to a mixture of 3-{4-[1-(3-hydroxy-6-methyl-pyridin-2-ylmethyl)-4,6-dimethyl-1H-benzoimidazol-2-ylamino]-piperidin-1-yl}-propionic acid ethyl ester (**i-1**; 0.00009 mol; melting point: 172°C; in tetrahydrofuran (10ml) under $N_2$ flow. The mixture was stirred at 5°C for 1 hour, then at room temperature for 3 hours. A minimum of $H_2O$ and ethylacetate were added. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was crystallized from 2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.026g of 2-{2-[1-(3-hydroxy-propyl)-piperidin-4-ylamino]-4,6-dimethyl-benzoimidazol-1-ylmethyl}-6-methyl-pyridin-3-ol (**i-2**; 68%, melting point: 209°C).

A mixture of **i-2** (0.0001 mol) and $CH_2Cl_2$ (15ml) was cooled in a bath of ice. $SOCl_2$ (0.0005 mol) was added drop wise. The mixture was stirred at 5°C for 1 hour, then at room temperature for 12 hours. $SOCl_2$ (0.0005 mol) was added. The mixture was stirred at room temperature for 4 hours. The solvent was evaporated till dryness, yielding 0.06g of intermediate **i-3** (100%). This product was used directly in the next reaction step.

A mixture of **i-3** (0.0001 mol), morpholine (0.0003 mol) and $K_2CO_3$ (0.0011 mol) in $CH_3CN$ (15ml) was stirred at 70°C for 6 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2/H_2O$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (0.06g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 88/11/1; 5µm). The pure fractions were collected and the solvent was evaporated, yielding 0.016g of 2-[4,6-dimethyl-2-(2-morpholin-4-ylpropyl-piperidin-4-ylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (18%, melting point: 223°C).

Scheme J




A mixture of **j-1** (0.166 mol) and urea (0.199 mol) in xylene (300 ml) was stirred under reflux for 12 hours. The reaction was cooled down to room temperature. The precipitate was filtered off, rinsed with xylene and diisopropylether, and then dried, yielding 32g of intermediate **j-2** (93%, melting point: > 260°C).

A mixture of **j-2** (0.073 mol) in $POCl_3$ (150 ml) was stirred at 100°C. HCl conc. (around 1.5 ml) was added drop wise very carefully until the dissolution of **j-2**. The mixture was stirred at 120°C for 6 hours. The solvent was evaporated till dryness. The residue was taken-up in $H_2O$/ice, basified with $K_2CO_3$ (powder) and extracted with ethylacetate + 10% methanol. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness, yielding 13.5 g of intermediate **j-3** (83%, melting point: 178°C).

A mixture of **j-3** (0.051 mmol) and **j-4** (0.056 mol) was stirred at 160°C for 2 hours. The residue was taken-up in $CH_2Cl_2/H_2O$ and basified with a 10% solution of $K_2CO_3$ in water. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent

was evaporated till dryness. The residue was purified by column chromatography over silica gel (eluent: $CH_2Cl_2$/methanol/$NH_4OH$ 95/5/0.5). The pure fractions were collected and the solvent was evaporated, yielding 15.3 g of intermediate **j-5** (79%).

A mixture of **j-5** (0.0396 mol), **j-6** (0.059 mol) and $K_2CO_3$ (0.1584 mol) in $CH_3CN$ (180ml) was stirred and refluxed for 12 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2$. The organic layer was washed with $H_2O$, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (20g) was purified by column chromatography over silica gel (eluent: Toluene/2-propanol/ $NH_4OH$ 85/15/1; 20-45μm). Two fractions were collected and the solvent was evaporated, yielding 5.3g of fraction 1 (27%) and 6.3g of fraction 2 (32%). Fraction 1 was crystallized twice in 2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 4.9g of intermediate **j-7** (25%, melting point: 179°C).

$LiAlH_4$ (0.009 mol) was added portion wise to a mixture of **j-7** (0.003 mol) in tetrahydrofuran (60 ml) at 5°C under $N_2$ flow. The reaction was stirred at 5°C for 1 hour and then at room temperature for 12 hours. Ethylacetate and $H_2O$ were added carefully and the aqueous layer was saturated with $K_2CO_3$ (powder). The organic layer was separated, dried (over $MgSO_4$) and then filtered over celite. The filtrate was evaporated till dryness, yielding 1.3 g of intermediate **j-8** (97%). The crude product was used directly in the next reaction step.

A mixture of **j-8** (0.0028 mol) and Pd/C 10% (2.5g) in $CH_3OH$ (40ml) was hydrogenated at 40°C for 12 hours under an 8 bar pressure, then filtered over celite. Celite was washed with a solution of $CH_3OH$/tetrahydrofuran (50/50). The filtrate was evaporated till dryness, yielding 1.8g of intermediate **j-9** (95%, melting point: 260°C).

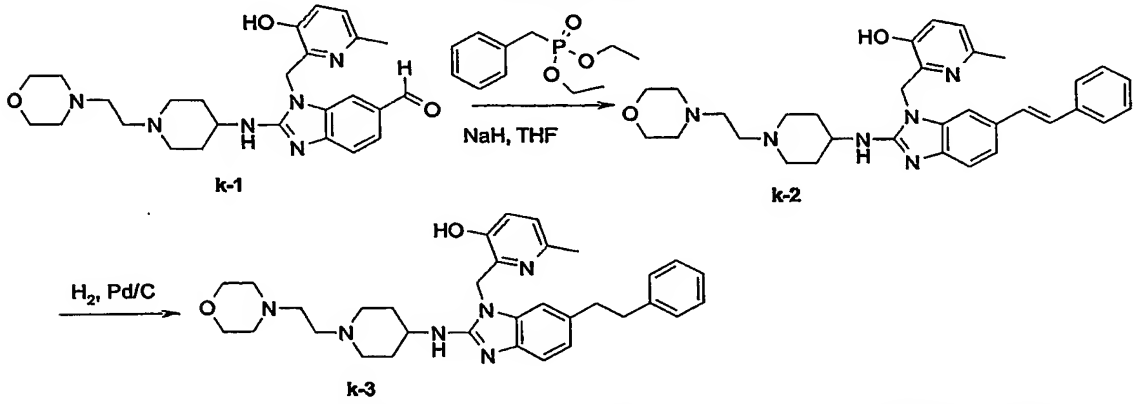
A mixture of **j-9** (0.0027 mol), N-(2-chloro-ethyl)-morpholine, HCl (0.0032 mol) and triethylamine (0.0067 mol) in dimethylformamide (40ml) was stirred at 50°C for 48 hours, poured out into ice water and extracted 3 times with $CH_2Cl_2$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was purified by column chromatography over silica gel (eluent: $CH_2Cl_2$/$CH_3OH$/$NH_4OH$; 85/14/1; 35-70μm). The pure fractions were collected and the solvent was evaporated. The residue was taken up in 2-propanone/diisopropylether. The precipitate was filtered off and dried, yielding 0.8g of intermediate **j-10** (61%, melting point: 147°C).

A mixture of **j-10** (0.0014 mol) and $MnO_2$ (1.6g) in $CH_2Cl_2$ (50ml) was stirred at room temperature for 12 hours, and then filtered over celite. The solvent of the filtrate was

evaporated till dryness. The residue was crystallized from 2-propanone/ diisopropylether. The precipitate was filtered off and dried, yielding 0.47g of intermediate **j-11** (67%, melting point: 136°C).

$CH_3CO_2H$ (0.3ml) was added at room temperature to a mixture of **j-11** (0.0005 mol), 3,5-dimethyl-aniline (0.0006 mol) and $NaBH_3CN$ (0.0006 mol) in $CH_3CN$ (30ml). The mixture was stirred at room temperature for 30 minutes. $CH_3CO_2H$ (0.3ml) was added. The mixture was stirred at room temperature for 6 hours. The solvent was evaporated till dryness. The residue was taken up in $CH_2Cl_2$. The organic layer was washed with a 10% solution of $K_2CO_3$ in water, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (0.26g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/1; 5µm). The pure fractions were collected and the solvent was evaporated. The residue (0.12g, 36%) was crystallized from $CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.07g of 2-{6-[(3,5-dimethyl-phenylamino)-methyl]-2-[2-(2-morpholin-4-yl-ethyl)-piperidin-4-ylamino]-benzoimidazol-1-ylmethyl}-6-methyl-pyridin-3-ol (21%, melting point: 150°C).

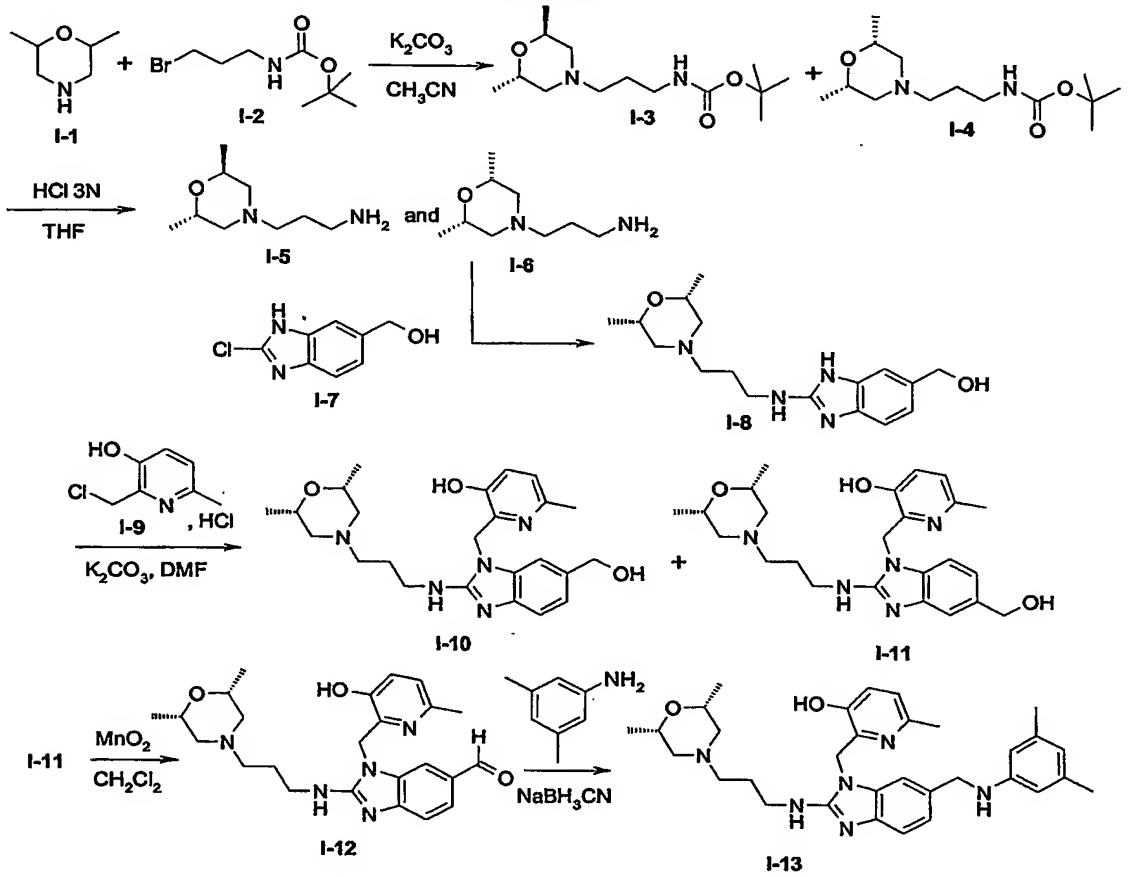
Scheme K




Benzyl-diethylphosphonate (0.0019 mol) was added to a mixture of NaH (0.0037 mol) in tetrahydrofuran (15ml) at 5°C under $N_2$ flow. The mixture was stirred at 5°C for 30 minutes. A solution of **k-1** (0.0006 mol) in tetrahydrofuran (10ml) was added drop wise. The mixture was stirred at 5°C for 1 hour, then at room temperature for 12 hours. $H_2O$ was added. The mixture was extracted with ethylacetate. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was crystallized from $CH_3OH$. The precipitate was filtered off and dried, yielding 0.13g of 6-methyl-2-{2-[2-(2-morpholin-4-yl-ethyl)-piperidin-4-ylamino]-6-styryl-benzoimidazol-1-ylmethyl}-pyridin-3-ol (**k-2**; 37%, melting point: 224°C).

A mixture of **k-2** (0.0002 mol) and Pd/C 10% (0.035g) in $CH_3OH$ (5ml) and tetrahydrofuran (5ml) was hydrogenated at room temperature for 6 hours under a 8 bar pressure, and then filtered over celite. Celite was washed with $H_2O$. The filtrate was evaporated till dryness. The residue was taken up in 2-propanone. The precipitate was filtered, washed with $H_2O$ and dried, yielding 0.08g of 6-methyl-2-{2-[2-(2-morpholin-4-yl-ethyl)-piperidin-4-ylamino]-6-phenethyl-benzoimidazol-1-ylmethyl}-pyridin-3-ol (72%, melting point: 159°C).

Scheme L




A mixture of **I-1** (mixture cis + trans) (0.0379 mol), **I-2** (0.0416 mol) and $K_2CO_3$ (0.1136 mol) was stirred at 80°C for 12 hours. $H_2O$ was added. The mixture was extracted with $CH_2Cl_2$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated. The residue (10g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 97/3/0.1; 35-70µm). Two fractions were collected and the solvent was evaporated, yielding 3g of intermediate **I-3** (trans) (29%) and 7.3g of intermediate **I-4** (cis) (71%).

A mixture of **I-4** (0.0279 mol) in a 3N solution of HCl in water (50ml) and tetrahydrofuran (50ml) was stirred at room temperature for 12 hours. $K_2CO_3$ (powder)

was added. $CH_2Cl_2$ was added. The aqueous layer was saturated with $K_2CO_3$ (powder). The mixture was extracted with $CH_2Cl_2$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated, yielding 4.39g of intermediate **I-6** (93%). Analogously, **I-5** was prepared.

A mixture of **I-7** (0.0085 mol) and **I-6** (0.0255 mol) was stirred at 120°C for 4 hours. A 10% solution of $K_2CO_3$ in water was added. The aqueous layer was saturated with $K_2CO_3$ (powder). The mixture was extracted with $CH_2Cl_2$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated. The residue (4.1g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/1; 15-40µm). The pure fractions were collected and the solvent was evaporated, yielding 1.6g of intermediate **I-8** (59%).

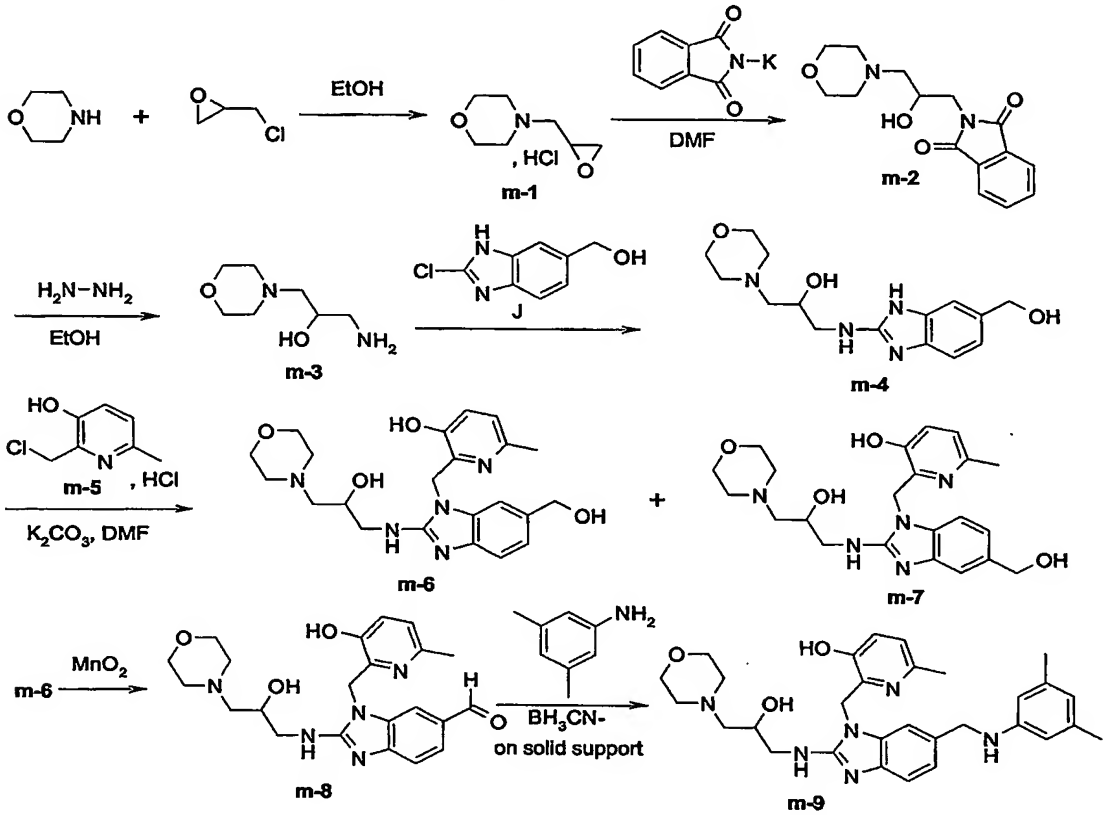
A mixture of **I-8** (0.0048 mol), **C** (0.0058 mol) and $K_2CO_3$ (0.0145 mol) in dimethylformamide (30ml) was stirred at room temperature for 24 hours, poured out into $H_2O$, saturated with $K_2CO_3$ (powder) and extracted with $CH_2Cl_2/CH_3OH$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue (3.3g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/0.5; 15-40µm). Two fractions were collected and the solvent was evaporated, yielding 0.55g of intermediate **I-10** (26%) and 0.36g of intermediate **I-11** (17%). A small fraction of intermediate **I-10** was crystallized from 2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.04g (melting point: 199°C). A small fraction of intermediate **I-11** was crystallized from 2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.04g (melting point: 227°C).

A mixture of **I-11** (0.0011 mol) and $MnO_2$ (1g) in $CH_2Cl_2$ (50ml) and $CH_3OH$ (3ml) was stirred at room temperature for 12 hours, and then filtered over celite. Celite was washed with $H_2O$. The filtrate was evaporated till dryness, yielding 0.5g of intermediate **I-12** (100%). The crude product was used directly in the next reaction step.

$CH_3CO_2H$ (0.25ml) was added to a mixture of **I-12** (0.0005 mol), 3,5-dimethyl-aniline (0.0006 mol) and $NaBH_3CN$ (0.0006 mol) in $CH_2Cl_2$ (30ml). The mixture was stirred at room temperature for 12 hours. A 10% solution of $K_2CO_3$ in water was added. The mixture was saturated with $K_2CO_3$ (powder). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 95/5/0.1; 35-70µm). The pure fractions were collected and the solvent was evaporated.

The residue (0.25g, 80%) was crystallized from 2-propanone/$CH_3CN$/diisopropylether. The precipitate was filtered off and dried, yielding 0.183g of 2-{2-[3-(2,6-dimethyl-morpholin-4-yl)-propylamino]-6-[(3,5-dimethyl-phenylamino)-methyl]-benzoimidazol-1-ylmethyl}-6-methyl-pyridin-3-ol (59%, melting point: 192°C).

Scheme M




A mixture of morpholine (0.0116 mol), epichlorohydrin (0.0116 mol) in ethanol (30 ml) was stirred at room temperature for 24 hours. The solvent was evaporated till dryness, yielding 2.08g of intermediate **m-1** (100%). The crude product was used directly in the next reaction step.

A mixture of **m-1** (0.0116 mol), potassium phthalimide (0.01276 mol) in dimethyl-formamide (25 ml) was stirred under reflux for 4 hours. The solvent was evaporated. The residue was taken up in $CH_2Cl_2$ and washed with $H_2O$. The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness, yielding 3.4g of intermediate **m-2** (100%). The crude product was used directly in the next reaction step.

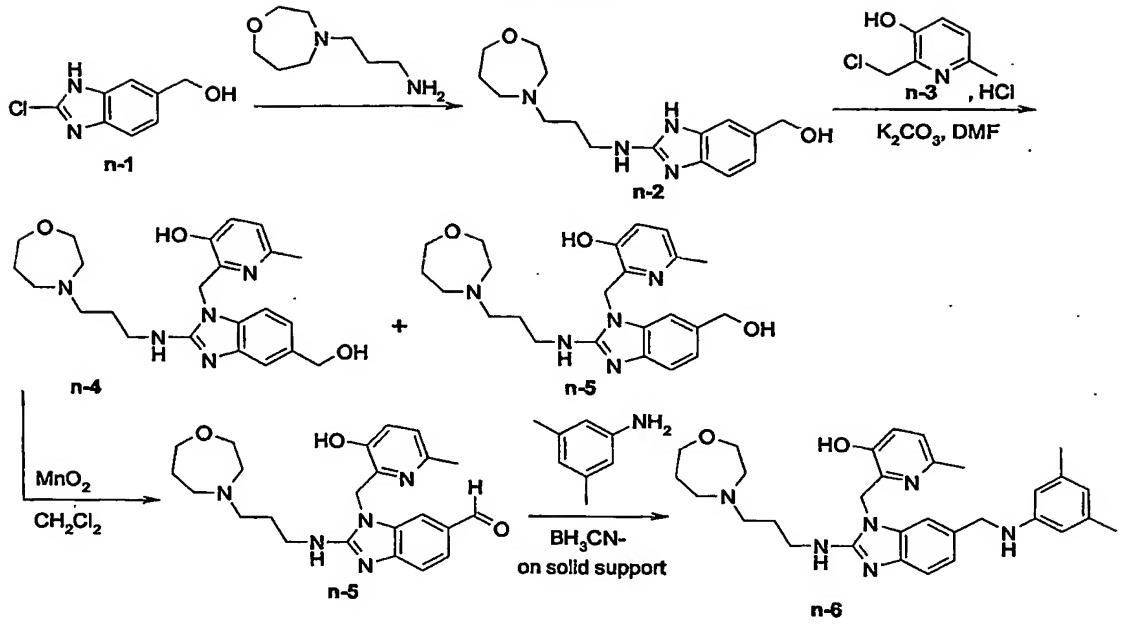
A mixture of **m-2** (0.116 mol) and hydrazine (15 ml) in ethanol (350 ml) was stirred at 80°C for 1 hour. The reaction was cooled down to room temperature. The precipitate

was filtered off and rinsed with ethanol and $CH_2Cl_2$. A 10% solution of $K_2CO_3$ in water was added. The aqueous layer was saturated with $K_2CO_3$ (powder) and extracted with $CH_2Cl_2/CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness, yielding 14.8g of intermediate **m-3** (80%). The crude product was used directly in the next reaction step.

Intermediate **m-5** was prepared in an analogous way to the procedure described for intermediate **l-8**. Intermediates **m-7** (2g; 31%, melting point: 184°C) and **m-8** (2.1g; 33%, melting point: 208°C) were prepared in an analogous way to the procedure described for preparing **l-10** and **l-11**. These compounds have been synthesized according to the procedure described for intermediates **J1** and **K1** and. Intermediate **m-9** (0.77g; 77%, melting point: 152°C) was prepared in an analogous way to the procedure described for intermediate **l-12**.

$CH_3CO_2H$ (0.2ml) was added at room temperature to a mixture of **m-9** (0.00047 mol), 3,5-dimethyl-aniline (0.00056 mol) and $BH_3CN$- on solid support (0.000705 mol) in $CH_3OH$ (10ml). The mixture was stirred at room temperature for 18 hours. The solid support was filtered off, rinsed with $CH_3OH$ and the filtrate was concentrated. The residue was taken up with a 10% solution of $K_2CO_3$ in water. The aqueous layer was saturated with $K_2CO_3$ (powder) and extracted with $CH_2Cl_2/CH_3OH$ (95/5). The organic layer was separated, dried (over $MgSO_4$), filtered and the solvent was evaporated till dryness. The residue was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 95/5/0.1; 35-70μm). The pure fractions were collected and the solvent was evaporated. The residue (0.2g) was crystallized from 2-propanone/ diisopropylether. The precipitate was filtered off and dried, yielding 0.154g of 2-[6-[(3,5-dimethyl-phenylamino)-methyl]-2-(2-hydroxy-3-morpholin-4-yl-propyl-amino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (**m-10**; 62%, melting point: 198°C).

Scheme N

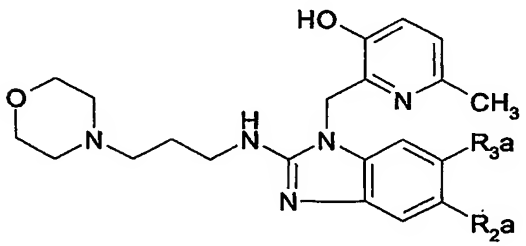


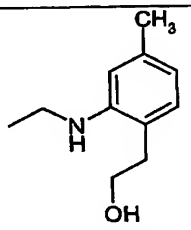
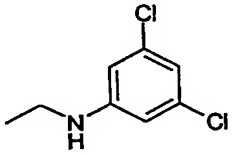
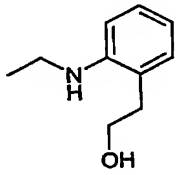
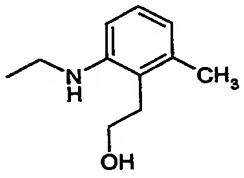
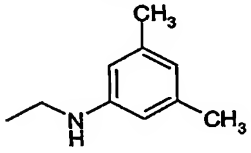
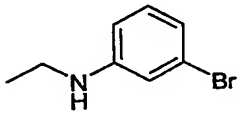
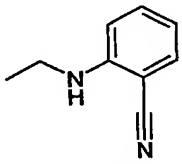
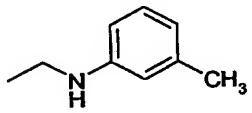
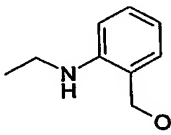
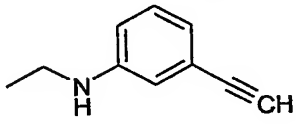

Intermediate **n-2** was prepared in an analogous way to the procedure described for intermediate **l-8**. Intermediates **n-4** (0.28g; 28%) and **n-5** (0.025g; 26%) were prepared in an analogous way to the procedure described for intermediate **l-10** and **l-11**. Intermediate **n-6** (0.020g; 80%) was prepared in an analogous way to the procedure described for intermediate **l-12**.

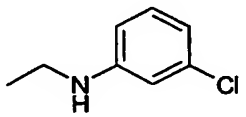
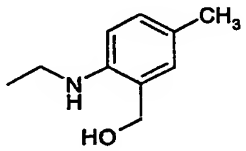
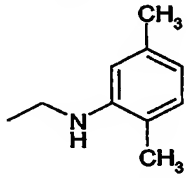
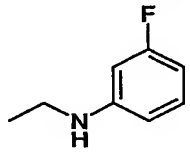
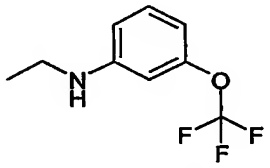
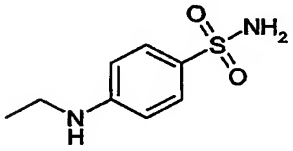
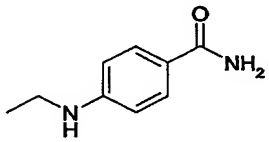
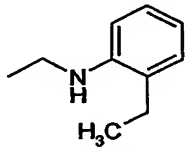
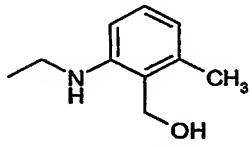
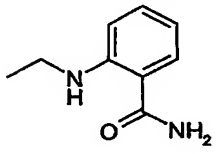
2-[5-[(3,5-Dimethyl-phenylamino)-methyl]-2-(3-[1,4]oxazepan-4-yl-propylamino)-benzoimidazol-1-ylmethyl]-6-methyl-pyridin-3-ol (0.007g; 28%) was prepared in an analogous way to the procedure described for compound **m-10**.

The following tables list compounds that were prepared according to any one of the above examples.

Table 1



| Comp. No. | R2a | R3a | Activity category | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|
| 1 | H |  | A | 545 | 199°C | D |
| 2 | H |  | A | 539-543 | 206°C | C |
| 3 | H |  | A | 531 | 140°C | D |
| 4 | H |  | A | 545 | 208°C | D |
| 5 | H |  | A | 515 | 199°C | C |
| 6 | H |  | A | 565-567 | 205°C | C |
| 7 | H |  | A | 512 | 217°C | D |
| 8 | H |  | A | 501 | 195°C | C |
| 9 | H |  | A | 517 | 130°C | D |
| 10 | H |  | A | 511 | 186°C | C |

| Comp. No. | R2a | R3a | Activity category | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|
| 11 | H |  | A | 522-524 | 212°C | C |
| 12 | H |  | A | 531 | 131°C | D |
| 13 | H |  | A | 515 | 209°C | D |
| 14 | H |  | A | 505 | 210°C | C |
| 15 | H |  | A | 571 | 163°C | D |
| 16 | H |  | A | 566 | > 260°C | D |
| 17 | H |  | A | 530 | 175°C | D |
| 18 | H |  | A | 515 | 210°C | D |
| 19 | H |  | A | 531 | 231°C | D |
| 20 | H |  | A | 530 | 145°C | D |

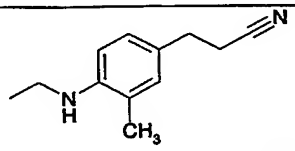
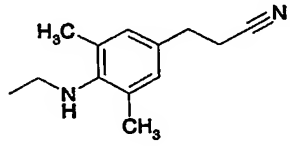
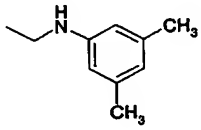
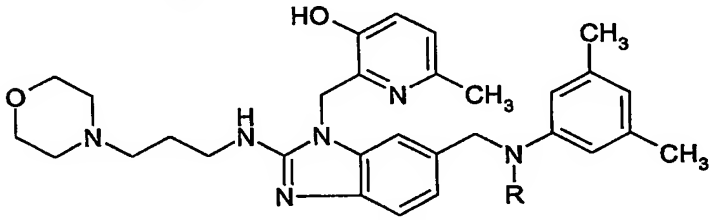
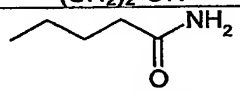
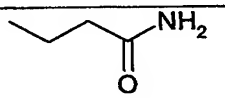
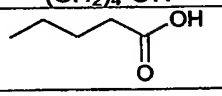
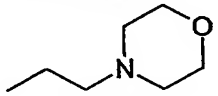
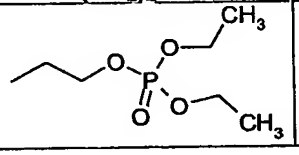
| Comp. No. | R2a | R3a | Activity category | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|
| 21 | H |  | A | 552 | 150°C | D |
| 22 | H |  | A | 568 | 114°C | E |
| 23 |  | H | A | 515 | 176°C | C |
| 24 | H | $-CH_2-OH$ | A | 412 | 192°C | A |
| 25 | $-CH_2-OH$ | H | C | 412 | 134°C | A |

Table 2 : compounds prepared according to synthesis scheme F



| Comp. No. | R | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|
| 26 | $-(CH_2)_2-OH$ | A | 559 | 180°C |
| 27 |  | A | 600 | 170°C |
| 28 |  | A | 586 | 138°C |
| 29 | $-(CH_2)_4-OH$ | A | 587 | 170°C |
| 30 |  | A | $MH^-$ = 599 | 121°C |
| 31 | $-(CH_2)_3-OH$ | A | 573 | 197°C |
| 32 | $-(CH_2)_5-OH$ | A | 601 | 120°C |
| 33 |  | A | 628 | 169°C |
| 34 | $-(CH_2)_2-NH_2$ | A | 558 | 196°C |
| 35 |  | A | 695 | 152°C |

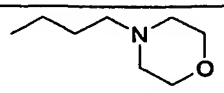
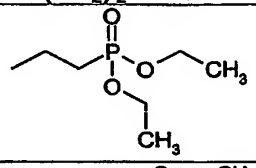
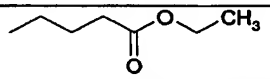
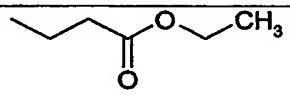
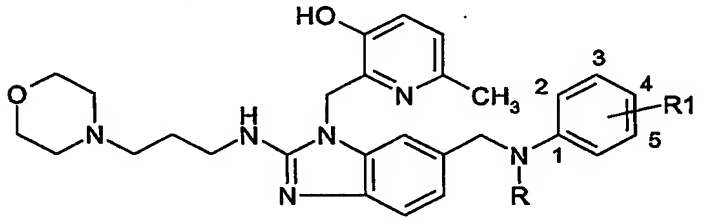
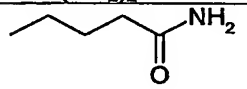
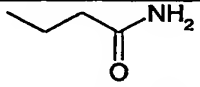
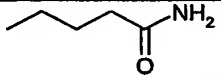
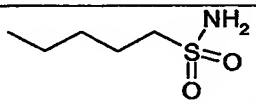
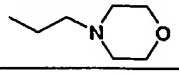
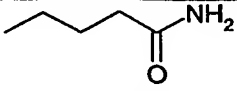
| Comp. No. | R | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|
| 36 |  | A | 642 | 169°C |
| 37 | $-(CH_2)_2-COOH$ | A | $MH^-$ = 585 | 128°C |
| 38 |  | A | 679 | 175°C |
| 39 |  | A | 629 | 130°C |
| 40 |  | A | 615 | 136°C |

Table 3 : compounds prepared according to synthesis scheme F



| Comp. No. | R | R1 | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|---|
| 41 | $-(CH_2)_2-OH$ | 3-Br | A | 609-611 | 210°C |
| 42 | $-(CH_2)_2-OH$ | 5-$CH_3$ | A | 545 | 205°C |
| 43 |  | 3-$CH_3$ | A | 586 | 139°C |
| 44 | $-(CH_2)_2-OH$ | 4-CN | A | 556 | 195°C |
| 45 |  | 3-$CH_3$ | A | 572 | 128°C |
| 46 |  | 5-Br | A | 650-652 | 180°C |
| 47 |  | 5-$CH_3$ | A | 636 | 140°C |
| 48 | $-(CH_2)_4-OH$ | 3-$CH_3$ | A | 573 | 169°C |
| 49 | $-(CH_2)_3-OH$ | 3-$CH_3$ | A | 559 | 109°C |
| 50 |  | 3-$CH_3$ | A | 614 | 153°C |
| 51 | $-(CH_2)_3-OH$ | 5-Br | A | 623-625 | 120°C |
| 52 |  | 4-CN | A | 597 | 170°C |

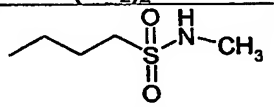
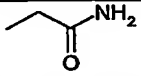
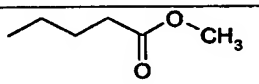
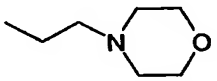
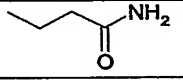
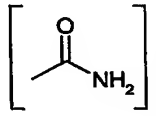
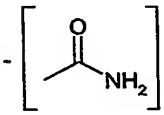
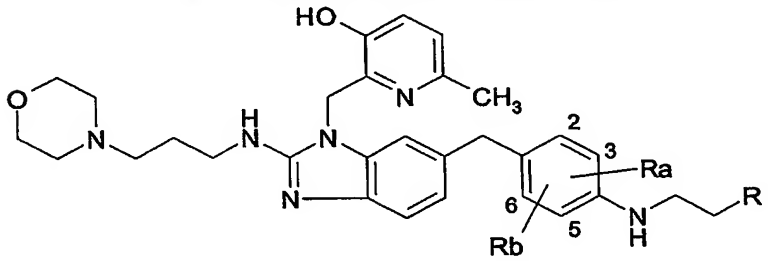
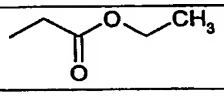
| Comp. No. | R | R1 | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|---|
| 53 | $-(CH_2)_2-OH$ | H | A | 531 | 190°C |
| 54 |  | 5-$CH_3$ | A | 636 | 125°C |
| 55 | $-(CH_2)_2-OH$ | 3-[-C≡CH] | A | 555 | 186°C |
| 56 | $-(CH_2)_2-N(CH_3)_2$ | 3-$CH_3$ | A | 572 | 172°C |
| 57 |  | 2-[$-(CH_2)_2-OH$] | A | 588 | 175°C |
| 58 |  | 3-$CH_3$ | A | 601 | 150°C |
| 59 |  | 6-[$-(CH_2)_2-OH$] | A | 644 | 146°C |
| 60 |  | 3-[$-(CH_2)_2-OH$] | A | 602 | 124°C |
| 61 | $-(CH_2)_2-OH$ | 4-  | A | 574 | 130°C |
| 62 | phenyl | 4-OH | A | 579 | 175°C |
| 63 | $-(CH_2)_2-OH$ | 6-[$-(CH_2)_2-OH$] | A | 575 | 165°C |
| 64 | H | 6-[$-CH_2-NH_2$] | A | 516 | 116°C |
| 65 | phenyl | 3-OH | A | 579 | 135°C |
| 66 | $-(CH_2)_2-OH$ | 6-$CH_3$ | A | 545 | 165°C |
| 67 | $-(CH_2)_2-OH$ | 2-  | A | 574 | 145°C |

Table 4 : compounds prepared according to synthesis scheme F



| Comp. No. | R | Ra | Rb | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|---|---|
| 68 |  | 2-$CH_3$ | 6-$CH_3$ | A | 629 | 164°C |
| 69 | OH | 3-$CH_3$ | H | A | 545 | 190°C |

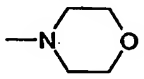
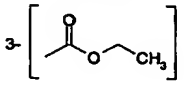
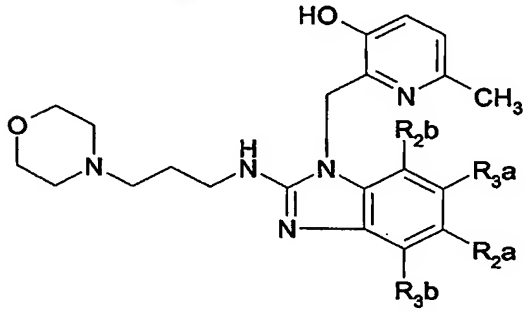
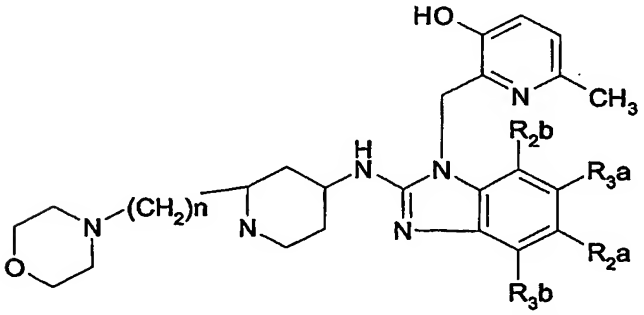
| Comp. No. | R | Ra | Rb | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|---|---|
| 70 | OH | 3-$OCH_3$ | H | A | 561 | 170°C |
| 71 | -$(CH_2)_2$-OH | 6-$CH_3$ | H | A | 573 | |
| 72 | OH | 2-$CH_3$ | 6-$CH_3$ | A | 559 | 162°C |
| 73 |  | 2-$CH_3$ | 6-$CH_3$ | A | 628 | 158°C |
| 74 | -$N(CH_3)_2$ | 2-$CH_3$ | 6-$CH_3$ | A | 586 | 140°C |
| 75 | OH |  | H | A | 603 | 150°C |
| 76 | -$(CH_2)_2$-OH | 2-$CH_3$ | 6-$CH_3$ | A | 587 | 156°C / HCl |

Table 5 : compounds prepared according to synthesis scheme G



| Comp. No. | R3b | R2a | R3a | R2b | Activity category | Mass (MH+) | Melting point/salt |
|---|---|---|---|---|---|---|---|
| 77 | -$CH_3$ | H | -$CH_3$ | H | A | 410 | 228°C |
| 78 | H | H | H | H | B | 382 | 203°C |
| 79 | H | -$CH_3$ | H | -$CH_3$ | C | 410 | 234°C |

Table 6 :



| Comp. No. | n | R3b | R2a | R3a | R2b | Activity | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|---|---|---|
| 80 | 3 | -$CH_3$ | H | -$CH_3$ | H | A | 493 | 223°C | I |

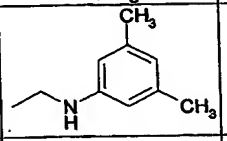
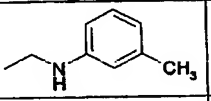
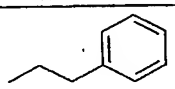
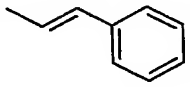
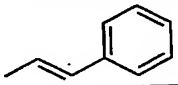
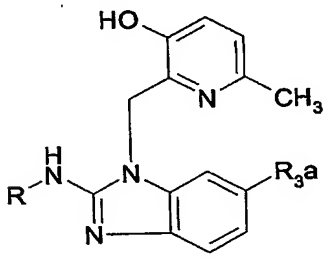
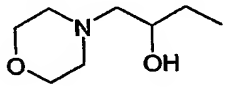
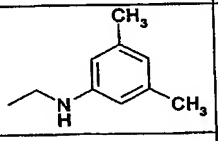
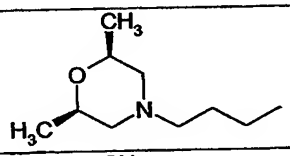
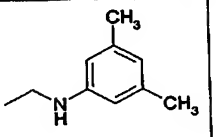
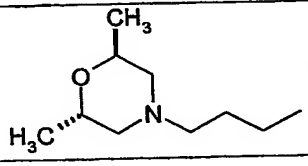
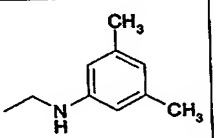
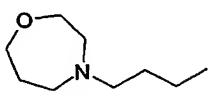
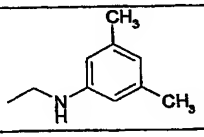
| Comp. No. | n | R3b | R2a | R3a | R2b | Activity | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|---|---|---|
| 81 | 2 | $-CH_3$ | H | $-CH_3$ | H | A | 479 | 226°C | H |
| 82 | 2 | H | H |  | H | A | 584 | 150°C | J |
| 83 | 2 | H | H |  | H | A | 570 | 130°C | J |
| 84 | 2 | H | H |  | H | B | 555 | 159°C | K |
| 85 | 2 | $-CH_3$ | H | H | H | B | 465 | 238°C | H |
| 86 | 2 | H |  | H | H | B | 553 | 225°C | K |
| 87 | 2 | H | H | $-CH_2-OH$ | H | B | 481 | 147°C | J |
| 88 | 2 | H | H |  | H | B | 553 | 224°C | K |
| 89 | 2 | H | $-CH_3$ | H | $-CH_3$ | B | 479 | 226°C | H |

Table 7



| Comp. No. | R | R3a | Activity category | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|
| 90 |  |  | A | 531 | 198°C | M |
| 91 |  |  | A | 543 | 192°C | L |
| 92 |  |  | A | 543 | 169°C | L |
| 93 |  |  | B | 529 | – | N |

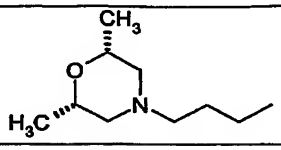
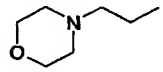
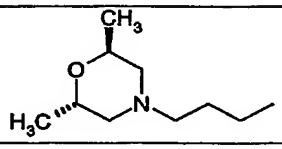
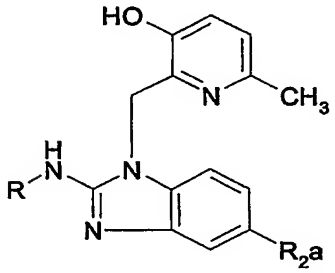
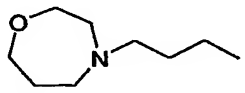
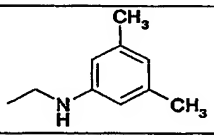
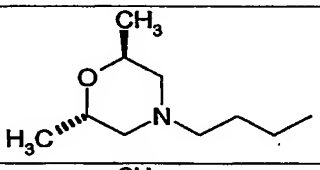
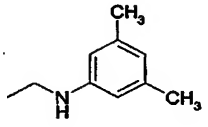
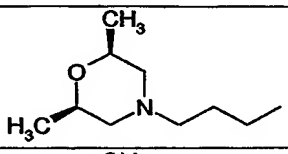
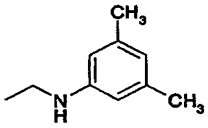
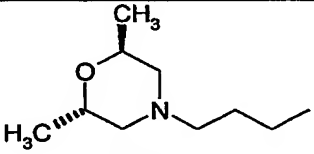
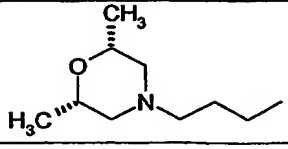
| Comp. No. | R | R3a | Activity category | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|
| 94 |  | -$CH_2$-OH | B | 440 | 199°C | L |
| 95 |  | H | C | 410 | 205°C | G |
| 96 |  | -$CH_2$-OH | C | 440 | 202°C | L |

Table 8



| Comp. No. | R | R2a | Activity | Mass (MH+) | Melting point/salt | Synthesis scheme |
|---|---|---|---|---|---|---|
| 97 |  |  | A | 529 | 198°C | N |
| 98 |  |  | B | 543 | 209°C | L |
| 99 |  |  | A | 543 | 169°C | L |
| 100 |  | -$CH_2$-OH | C | 440 | 212 | L |
| 101 |  | -$CH_2$-OH | C | 440 | 227°C | L |

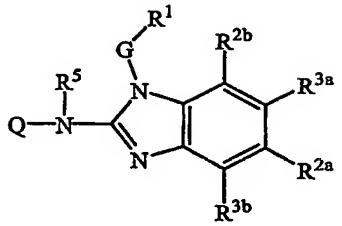
***B. In vitro screening for activity against Respiratory Syncytial Virus.***

The percent protection against cytopathology caused by viruses (antiviral activity or $IC_{50}$) achieved by tested compounds and their cytotoxicity ($CC_{50}$) are both calculated from dose-response curves. The selectivity of the antiviral effect is represented by the selectivity index (SI), calculated by dividing the $CC_{50}$ (cytotoxic dose for 50% of the cells) by the $IC_{50}$ (antiviral activity for 50 % of the cells). The tables in the above experimental part list the category to which each of the prepared compounds belong : Compounds belonging to activity category "A" have an $pIC_{50}$ (-log of $IC_{50}$ when expressed in molar units) equal to or more than 7. Compounds belonging to activity category "B" have a pIC50 value between 6 and 7. Compounds belonging to activity category "C" have a pIC50 value equal to or below 6.

Automated tetrazolium-based colorimetric assays were used for determination of $IC_{50}$ and $CC_{50}$ of test compounds. Flat-bottom, 96-well plastic microtiter trays were filled with 180 µl of Eagle's Basal Medium, supplemented with 5 % FCS (0% for FLU) and 20 mM Hepes buffer. Subsequently, stock solutions (7.8 x final test concentration) of compounds were added in 45 µl volumes to a series of triplicate wells so as to allow simultaneous evaluation of their effects on virus- and mock-infected cells. Five five-fold dilutions were made directly in the microtiter trays using a robot system. Untreated virus controls, and HeLa cell controls were included in each test. Approximately 100 $TCID_{50}$ of Respiratory Syncytial Virus was added to two of the three rows in a volume of 50 µl. The same volume of medium was added to the third row to measure the cytotoxicity of the compounds at the same concentrations as those used to measure the antiviral activity. After two hours of incubation, a suspension ($4 \times 10^5$ cells/ml) of HeLa cells was added to all wells in a volume of 50µl. The cultures were incubated at 37°C in a 5% $CO_2$ atmosphere. Seven days after infection the cytotoxicity and the antiviral activity was examined spectrophotometrically. To each well of the microtiter tray, 25 µl of a solution of MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) was added . The trays were further incubated at 37°C for 2 hours, after which the medium was removed from each cup. Solubilization of the formazan crystals was achieved by adding 100 µl 2-propanol. Complete dissolution of the formazan crystals were obtained after the trays have been placed on a plate shaker for 10 min. Finally, the absorbances were read in an eight-channel computer-controlled photometer (Multiskan MCC, Flow Laboratories) at two wavelengths (540 and 690 nm). The absorbance measured at 690 nm was automatically subtracted from the absorbance at 540 nm, so as to eliminate the effects of non-specific absorption.

Claims

1. A compound having the formula

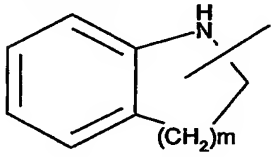


(I)

a prodrug, *N*-oxide, addition salt, quaternary amine, metal complex or stereochemically isomeric form thereof wherein

G is a direct bond or $C_{1-10}$alkanediyl optionally substituted with one or more substituents individually selected from the group of substituents consisting of hydroxy, $C_{1-6}$alkyloxy, $Ar^1C_{1-6}$alkyloxy, $C_{1-6}$alkylthio, $Ar^1C_{1-6}$alkylthio, $HO(-CH_2-CH_2-O)_n-$, $C_{1-6}$alkyloxy$(-CH_2-CH_2-O)_n-$ or $Ar^1C_{1-6}$alkyloxy$(-CH_2-CH_2-O)_n-$;

$R^1$ is $Ar^1$ or a monocyclic or bicyclic heterocycle being selected from piperidinyl, piperazinyl, pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, furanyl, tetrahydrofuranyl, thienyl, pyrrolyl, thiazolyl, oxazolyl, imidazolyl, isothiazolyl, pyrazolyl, isoxazolyl, oxadiazolyl, quinolinyl, quinoxalinyl, benzofuranyl, benzothienyl, benzimidazolyl, benzoxazolyl, benzthiazolyl, pyridopyridyl, naphthiridinyl, 1*H*-imidazo[4,5-b]pyridinyl, 3*H*-imidazo[4,5-b]pyridinyl, imidazo[1,2-a]pyridinyl, 2,3-dihydro-1,4-dioxino[2,3-b]pyridyl or a radical of formula



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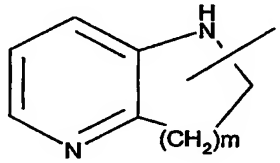

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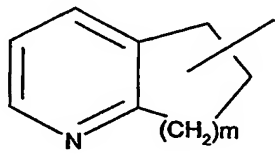

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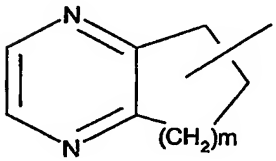

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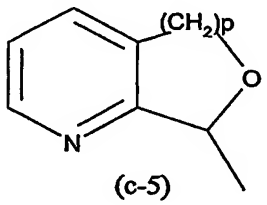

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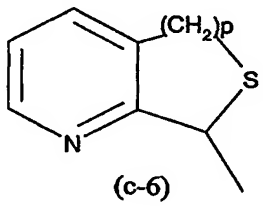

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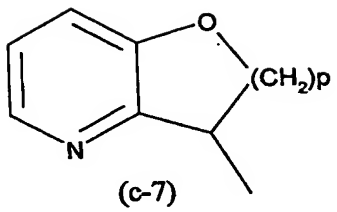
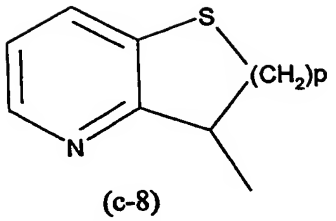



(c-7)




(c-8) ;

wherein each of said monocyclic or bicyclic heterocycles may optionally be substituted with 1 or where possible more, such as 2, 3, 4 or 5, substituents individually selected from the group of substituents consisting of halo, hydroxy, amino, cyano, carboxyl, $C_{1-6}$alkyl, $C_{1-6}$alkyloxy, $C_{1-6}$alkylthio, $C_{1-6}$alkyloxy$C_{1-6}$alkyl, $Ar^1$, $Ar^1C_{1-6}$alkyl, $Ar^1C_{1-6}$alkyloxy, hydroxy$C_{1-6}$alkyl, mono-or di($C_{1-6}$alkyl)amino, mono-or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyl, $C_{1-6}$alkylcarbonylamino, $C_{1-6}$alkyl-$SO_2$-$NR^{5c}$-, $Ar^1$-$SO_2$-$NR^{5c}$-, $C_{1-6}$alkyloxycarbonyl, -C(=O)-$NR^{5c}R^{5d}$, HO(-$CH_2$-$CH_2$-O)$_n$-, halo(-$CH_2$-$CH_2$-O)$_n$-, $C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$-, $Ar^1C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$- and mono-or di($C_{1-6}$alkyl)amino(-$CH_2$-$CH_2$-O)$_n$-;

each n independently is 1, 2, 3 or 4;

each m independently is 1 or 2;

each p independently is 1 or 2;

each t independently is 0, 1 or 2;

Q is $R^7$, pyrrolidinyl substituted with $R^7$, piperidinyl substituted with $R^7$ or homopiperidinyl substituted with $R^7$ wherein $R^7$ is $C_{1-6}$alkyl substituted with a heterocycle selected from the group consisting of morpholinyl, thiomorpholinyl, 1-oxo-thiomorpholinyl and 1,1-dioxothiomorpholinyl, wherein each of said heterocyle may be optionally substituted with one or two substituents selected from the group consisting of hydroxy, carboxyl, $C_{1-4}$alkyloxycarbonyl, aminocarbonyl, mono- or di($C_{1-4}$alkyl)-aminocarbonyl, $C_{1-4}$alkylcarbonylamino, aminosulfonyl and mono- or di($C_{1-4}$alkyl)-aminosulfonyl;

one of $R^{2a}$ and $R^{3a}$ is selected from halo, optionally mono- or polysubstituted $C_{1-6}$alkyl, optionally mono- or polysubstituted $C_{2-6}$alkenyl, nitro, hydroxy, $Ar^2$, $N(R^{4a}R^{4b})$, $N(R^{4a}R^{4b})$sulfonyl, $N(R^{4a}R^{4b})$carbonyl, $C_{1-6}$alkyloxy, $Ar^2$oxy, $Ar^2C_{1-6}$alkyloxy, carboxyl, $C_{1-6}$alkyloxycarbonyl, or -C(=Z)$Ar^2$; and the other one of $R^{2a}$ and $R^{3a}$ is hydrogen;

wherein

- =Z is =O, =CH-C(=O)-$NR^{5a}R^{5b}$, =$CH_2$, =CH-$C_{1-6}$alkyl, =N-OH or =N-O-$C_{1-6}$alkyl; and
- the optional substituents on $C_{1-6}$alkyl and $C_{2-6}$alkenyl can be the same or can be different relative to one another, and are each independently selected from the group of substituents consisting of hydroxy, cyano, halo, nitro, $N(R^{4a}R^{4b})$,

$N(R^{4a}R^{4b})$sulfonyl, Het, $Ar^2$, $C_{1-6}$alkyloxy, $C_{1-6}$alkyl-$S(=O)_t$, $Ar^2$oxy, $Ar^2$-$S(=O)_t$, $Ar^2C_{1-6}$alkyloxy, $Ar^2C_{1-6}$alkyl-$S(=O)_t$, Het-oxy, Het-$S(=O)_t$, $HetC_{1-6}$alkyloxy, $HetC_{1-6}$alkyl-$S(=O)_t$, carboxyl, $C_{1-6}$alkyloxycarbonyl and $-C(=Z)Ar^2$;

in case $R^{2a}$ is different from hydrogen then $R^{2b}$ is hydrogen, $C_{1-6}$alkyl or halogen and $R^{3b}$ is hydrogen;

in case $R^{3a}$ is different from hydrogen then $R^{3b}$ is hydrogen, $C_{1-6}$alkyl or halogen and $R^{2b}$ is hydrogen;

$R^{4a}$ and $R^{4b}$ can be the same or can be different relative to one another, and are each independently selected from the group of substituents consisting of hydrogen, $C_{1-6}$alkyl, $Ar^2$, $Ar^2C_{1-6}$alkyl, $C_{1-6}$alkylcarbonyl, $Ar^2$carbonyl, $Ar^2C_{1-6}$alkyl-carbonyl, $C_{1-6}$alkylsulfonyl, $Ar^2$sulfonyl, $Ar^2C_{1-6}$alkylsulfonyl, $C_{1-6}$alkyloxy-$C_{1-6}$alkyl, amino$C_{1-6}$alkyl, mono- or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, hydroxy-$C_{1-6}$alkyl, Het, Het-$C_{1-6}$alkyl, Het-carbonyl, Het-sulfonyl, $HetC_{1-6}$alkylsulfonyl and Het-$C_{1-6}$alkylcarbonyl;

$R^{5a}$ and $R^{5b}$ can be the same or can be different relative to one another, and are each independently hydrogen or $C_{1-6}$alkyl; or

$R^{5a}$ and $R^{5b}$ taken together may form a bivalent radical of formula $-(CH_2)_s-$ wherein s is 4 or 5;

$R^{5c}$ and $R^{5d}$ can be the same or can be different relative to one another, and are each independently hydrogen or $C_{1-6}$alkyl; or

$R^{5c}$ and $R^{5d}$ taken together may form a bivalent radical of formula $-(CH_2)_s-$ wherein s is 4 or 5;

$R^{6a}$ is hydrogen, $C_{1-6}$alkyl, $Ar^1$, $Ar^1C_{1-6}$alkyl, $C_{1-6}$alkylcarbonyl, $Ar^1$carbonyl, $Ar^1C_{1-6}$alkylcarbonyl, $C_{1-6}$alkylsulfonyl, $Ar^1$sulfonyl, $Ar^1C_{1-6}$alkylsulfonyl, $C_{1-6}$alkyloxy$C_{1-6}$alkyl, amino$C_{1-6}$alkyl, mono- or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, hydroxy$C_{1-6}$alkyl, Het, Het-$C_{1-6}$alkyl, Het-carbonyl, Het-sulfonyl, Het-$C_{1-6}$alkyl-carbonyl;

$R^{6b}$ is hydrogen, $C_{1-6}$alkyl, $Ar^1$ or $Ar^1C_{1-6}$alkyl;

$R^{6c}$ is $C_{1-6}$alkyl, $Ar^1$ or $Ar^1C_{1-6}$alkyl;

$Ar^1$ is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, $C_{1-6}$alkyl, hydroxy$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyl, and $C_{1-6}$alkyloxy;

$Ar^2$ is phenyl, phenyl annealed with a $C_{5-7}$cycloalkyl, or phenyl substituted with 1 or more, such as 2, 3, 4 or 5, substituents selected from halo, cyano, $C_{1-6}$alkyl, cyano$C_{1-6}$alkyl, cyano$C_{2-6}$alkenyl, $Ar^1$, $R^{6b}$-O-, $R^{6b}$-S-, $N(R^{6a}R^{6b})$, polyhalo$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyloxy, polyhalo$C_{1-6}$alkylthio, $R^{6c}$-C(=O)-, $R^{6b}$-O-C(=O)-, $N(R^{6a}R^{6b})$-C(=O)-, $R^{6b}$-O-$C_{1-6}$alkyl, $R^{6b}$-S-$C_{1-6}$alkyl,

$R^{6c}$-S(=O)$_2$-$C_{1-6}$alkyl, N($R^{6a}R^{6b}$)-$C_{1-6}$alkyl, $R^{6c}$-C(=O)-$C_{1-6}$alkyl, $R^{6b}$-O-C(=O)-$C_{1-6}$alkyl, N($R^{6a}R^{6b}$)-C(=O)-$C_{1-6}$alkyl, $R^{6c}$-C(=O)-$NR^{6b}$-, $R^{6c}$-C(=O)-O-, $R^{6c}$-C(=O)-$NR^{6b}$-$C_{1-6}$alkyl, $R^{6c}$-C(=O)-O-$C_{1-6}$alkyl, N($R^{6a}R^{6b}$)-S(=O)$_2$-;

Het is a heterocycle being selected from tetrahydrofuranyl, tetrahydrothienyl, pyrrolidinyl, pyrrolidinonyl, furanyl, thienyl, pyrrolyl, thiazolyl, oxazolyl, imidazolyl, isothiazolyl, pyrazolyl, isoxazolyl, oxadiazolyl, thiadiazolyl, piperidinyl, homopiperidinyl, piperazinyl, morpholinyl, pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, tetrahydroquinolinyl, quinolinyl, isoquinolinyl, benzodioxanyl, benzodioxolyl, indolinyl, indolyl, each of said heterocycle may optionally be substituted with oxo, amino, $Ar^1$, $C_{1-4}$alkyl, amino$C_{1-4}$alkyl, $Ar^1C_{1-4}$alkyl, mono- or di($C_{1-6}$alkyl)amino$C_{1-6}$alkyl, mono- or di($C_{1-6}$alkyl)amino.

2. A compound as claimed in claim 1 wherein G is $C_{1-10}$alkanediyl.

3. A compound as claim in claim 1 or 2 wherein $R^1$ is pyridyl optionally substituted with 1 or 2 substituents individually selected from the group of substituents consisting of halo, hydroxy, amino, cyano, carboxyl, $C_{1-6}$alkyl, $C_{1-6}$alkyloxy, $C_{1-6}$alkylthio, $C_{1-6}$alkyloxy$C_{1-6}$alkyl, $Ar^1$, $Ar^1C_{1-6}$alkyl, $Ar^1C_{1-6}$alkyloxy, hydroxy$C_{1-6}$alkyl, mono-or di($C_{1-6}$alkyl)amino, mono-or di($C_{1-6}$alkyl)amino-$C_{1-6}$alkyl, polyhalo$C_{1-6}$alkyl, $C_{1-6}$alkylcarbonylamino, $C_{1-6}$alkyl-$SO_2$-$NR^{5c}$-, $Ar^1$-$SO_2$-$NR^{5c}$-, $C_{1-6}$alkyloxycarbonyl, -C(=O)-$NR^{5c}R^{5d}$, HO(-$CH_2$-$CH_2$-O)$_n$-, halo(-$CH_2$-$CH_2$-O)$_n$-, $C_{1-6}$alkyloxy(-$CH_2$-$CH_2$-O)$_n$-, $Ar^1C_{1-6}$alkyloxy(-$CH_2$-$CH_2O$)$_n$- and mono-or di($C_{1-6}$alkyl)amino(-$CH_2$-$CH_2$-O)$_n$-.

4. A compound as claimed in any one of claim 1 to 3 wherein t is 2.

5. A compound as claimed in any one of claim 1 to 4 wherein the compound has the structure of the compound numbers 1 to 24, 26 to 77, 80 to 83, 90 to 92, 97 and 99 listed in tables 1 to 8.

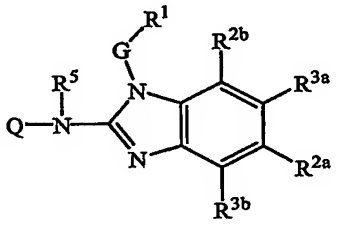
6. A compound as claimed in any one of claims 1 to 5 for use as a medicine.

7. A pharmaceutical compositioncomprising a pharmaceutically acceptable carrier, and as active ingredient a therapeutically effective amount of a compound as described in any one of claims 1 to 5.

## ABSTRACT

# MORPHOLINYL CONTAINING BENZIMIDAZOLES AS INHIBITORS OF RESPIRATORY SYNCYTIAL VIRUS REPLICATION

The present invention concerns morpholinyl containing benzimidazoles having inhibitory activity on the replication of the respiratory syncytial virus and having the formula



(I)

a prodrug, *N*-oxide, addition salt, quaternary amine, metal complex or stereochemically isomeric form thereof wherein G is a direct bond or optionally substituted $C_{1-10}$alkanediyl; $R^1$ is $Ar^1$ or a monocyclic or bicyclic heterocycle Q is $R^7$, pyrrolidinyl substituted with $R^7$, piperidinyl substituted with $R^7$ or homopiperidinyl substituted with $R^7$; one of $R^{2a}$ and $R^{3a}$ is selected from halo, optionally mono- or polysubstituted $C_{1-6}$alkyl, optionally mono- or polysubstituted $C_{2-6}$alkenyl, nitro, hydroxy, $Ar^2$, $N(R^{4a}R^{4b})$, $N(R^{4a}R^{4b})$sulfonyl, $N(R^{4a}R^{4b})$carbonyl, $C_{1-6}$alkyloxy, $Ar^2$oxy, $Ar^2C_{1-6}$alkyloxy, carboxyl, $C_{1-6}$alkyloxycarbonyl, or $-C(=Z)Ar^2$; and the other one of $R^{2a}$ and $R^{3a}$ is hydrogen; in case $R^{2a}$ is different from hydrogen then $R^{2b}$ is hydrogen, $C_{1-6}$alkyl or halogen and $R^{3b}$ is hydrogen; in case $R^{3a}$ is different from hydrogen then $R^{3b}$ is hydrogen, $C_{1-6}$alkyl or halogen and $R^{2b}$ is hydrogen. It further concerns their preparation and compositions comprising them, as well as their use as a medicine.

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